

University of Mumbai



No. AAMS_UGS/ICC/2022-23/ 126

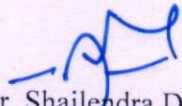
CIRCULAR :-

Attention of the Principals of the Affiliated Colleges and Directors of the recognized Institutions in Faculty of Science & Technology is invited to this office circular No.UG/146 of 2016-17 dated 9th November, 2016 relating to the revised syllabus of Master of Engineering (Chemical Engineering) (Sem.- I to IV) (CBCS) .

They are hereby informed that the recommendations made by the Ad-hoc Board of Studies in Chemical Engineering at its meeting held on 04th May, 2022 and subsequently passed in the Faculty and then by the Board of Deans at its meeting held on 5th July, 2022 vide item No. 6.38 (R) have been accepted by the Academic Council at its meeting held on 11th July, 2022 vide item No. 6.38 (R) and that in accordance therewith, the revised syllabus of M.E.(Chemical Engineering) (Sem.- I to IV) (CBCS) (REV-2022 Scheme) has been brought into force with effect from the academic year 2022-23. (The circular is available on the University's website www.mu.ac.in).

MUMBAI - 400 032

20th October, 2022


(Dr. Shailendra Deolankar)
I/c Registrar

To

The Principals of the Affiliated Colleges and Directors of the recognized Institutions in Faculty of Science & Technology.

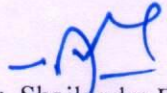
A.C/6.38(R)/11/07/2022

No. AAMS_UGS/ICC/ 2022-23/ 126

20th October, 2022

Copy forwarded with Compliments for information to:-

- 1) The Dean, Faculty of Science & Technology,
- 2) The Chairman, Ad-hoc Board of Studies in Chemical Engineering,
- 3) The Director, Board of Examinations and Evaluation,
- 4) The Director, Board of Students Development,
- 5) The Director, Department of Information & Communication Technology,
- 6) The Co-ordinator, MKCL.


(Dr. Shailendra Deolankar)
I/c Registrar

Desktop/Circular of Engineering/Priya

Copy to :-

- 1. The Deputy Registrar, Academic Authorities Meetings and Services (AAMS),**
- 2. The Deputy Registrar, College Affiliations & Development Department (CAD),**
- 3. The Deputy Registrar, (Admissions, Enrolment, Eligibility and Migration Department (AEM),**
- 4. The Deputy Registrar, Research Administration & Promotion Cell (RAPC),**
- 5. The Deputy Registrar, Executive Authorities Section (EA),**
- 6. The Deputy Registrar, PRO, Fort, (Publication Section),**
- 7. The Deputy Registrar, (Special Cell),**
- 8. The Deputy Registrar, Fort/ Vidyanagari Administration Department (FAD) (VAD), Record Section,**
- 9. The Director, Institute of Distance and Open Learning (IDOL Admin), Vidyanagari,**

They are requested to treat this as action taken report on the concerned resolution adopted by the Academic Council referred to in the above circular and that on separate Action Taken Report will be sent in this connection.

- 1. P.A to Hon'ble Vice-Chancellor,**
- 2. P.A Pro-Vice-Chancellor,**
- 3. P.A to Registrar,**
- 4. All Deans of all Faculties,**
- 5. P.A to Finance & Account Officers, (F.& A.O),**
- 6. P.A to Director, Board of Examinations and Evaluation,**
- 7. P.A to Director, Innovation, Incubation and Linkages,**
- 8. P.A to Director, Board of Lifelong Learning and Extension (BLLE),**
- 9. The Director, Dept. of Information and Communication Technology (DICT) (CCF & UCC), Vidyanagari,**
- 10. The Director of Board of Student Development,**
- 11. The Director, Department of Students Welfare (DSD),**
- 12. All Deputy Registrar, Examination House,**
- 13. The Deputy Registrars, Finance & Accounts Section,**
- 14. The Assistant Registrar, Administrative sub-Campus Thane,**
- 15. The Assistant Registrar, School of Engg. & Applied Sciences, Kalyan,**
- 16. The Assistant Registrar, Ratnagiri sub-centre, Ratnagiri,**
- 17. The Assistant Registrar, Constituent Colleges Unit,**
- 18. BUCTU,**
- 19. The Receptionist,**
- 20. The Telephone Operator,**
- 21. The Secretary MUASA**

for information.

AC – 11 July, 2022

Item No. – 6.38 (R)

University of Mumbai



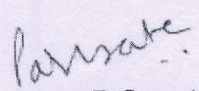
**Revised Syllabus for
M.E
Chemical Engineering
(Sem. - I to IV)
(Choice Based Credit System)**

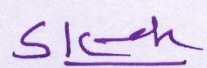
(With effect from the academic year 2022-23)

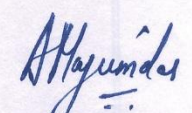
University of Mumbai



O: _____	Title of Course	M.E (Chemical Engineering)
O: _____	Eligibility	Passing Bachelor of Engineering as per the Ordinance 0.6264
R: _____	Passing Marks	45%
No. of years/Semesters:		2 Years/ 4 Semesters
Level:		P.G. / U.G./ Diploma / Certificate
Pattern:		Yearly / Semester
Status:		New / Revised
To be implemented from Academic Year:		With effect from Academic Year: 2022-23


Dr. Parag R Gogate
Chairman
of Ad-hoc Board of Studies in
Chemical Engineering


Dr. Suresh K. Ukarande
Associate Dean,
Faculty of Science and
Technology


Dr Anuradha Majumdar
Dean,
Faculty of Science and
Technology

Preamble

To meet the challenge of ensuring excellence in engineering education, the issue of quality needs to be addressed, debated and taken forward in a systematic manner. Accreditation is the principal means of quality assurance in higher education. The major emphasis of accreditation process is to measure the outcomes of the program that is being accredited. In line with this Faculty of Science and Technology (in particular Engineering) of University of Mumbai has taken a lead in incorporating philosophy of outcome-based education in the process of curriculum development.

Faculty resolved that course objectives and course outcomes are to be clearly defined for each course, so that all faculty members in affiliated institutes understand the depth and approach of course to be taught, which will enhance learner's learning process. Choice based Credit and grading system enables a much-required shift in focus from teacher-centric to learner-centric education since the workload estimated is based on the investment of time in learning and not in teaching. It also focuses on continuous evaluation which will enhance the quality of education. Credit assignment for courses is based on 15 weeks teaching learning process, however content of courses is to be taught in 13 weeks and remaining 2 weeks to be utilized for revision, guest lectures, coverage of content beyond syllabus etc.

There was a concern that the earlier revised curriculum more focused on providing information and knowledge across various domains of the said program, which led to heavily loading of students in terms of direct contact hours. In this regard, faculty of science and technology resolved that to minimize the burden of contact hours, total credits of entire program will be of 68, wherein focus is not only on providing knowledge but also on building skills, attitude and self-learning. Therefore, in the present curriculum skill-based laboratories are made mandatory across all disciplines of engineering in first and second semester of programs, which will definitely facilitate self-learning of students. The overall credits and approach of curriculum proposed in the present revision is in line with AICTE model curriculum.

The present curriculum will be implemented for Master of Engineering from the academic year 2022-23

Dr. S. K. Ukarande

Associate Dean

Faculty of Science and Technology

University of Mumbai

Dr Anuradha Muzumdar

Dean

Faculty of Science and Technology

University of Mumbai

Incorporation and Implementation of Online Contents from NPTEL/ Swayam Platform

The curriculum revision is mainly focused on knowledge component, skill based activities and project based activities. Self-learning opportunities are provided to learners. In the revision process this time in particular Revised syllabus of 'C' scheme wherever possible additional resource links of platforms such as NPTEL, Swayam are appropriately provided. In an earlier revision of curriculum in the year 2012 and 2016 in Revised scheme 'A' and 'B' respectively, efforts were made to use online contents more appropriately as additional learning materials to enhance learning of students.

In the current revision based on the recommendation of AICTE model curriculum overall credits are reduced to 68, to provide opportunity of self-learning to learner. Learners are now getting sufficient time for self-learning either through online courses or additional projects for enhancing their knowledge and skill sets.

The Principals/ HoD's/ Faculties of all the institute are required to motivate and encourage learners to use additional online resources available on platforms such as NPTEL/ Swayam. Learners can be advised to take up online courses, on successful completion they are required to submit certification for the same. This will definitely help learners to facilitate their enhanced learning based on their interest.

Dr. S. K. Ukarande

Associate Dean

Faculty of Science and Technology

University of Mumbai

Dr Anuradha Muzumdar

Dean

Faculty of Science and Technology

University of Mumbai

Preamble to the Revision of Syllabus in Chemical Engineering

Development in all fields including Chemical Engineering along with use of soft wares for process plant and process engineering, there is demand on academicians to upgrade the curriculum in Education. Choice based Credit and grading system enables a much-required shift in focus from teacher-centric to learner-centric education since the workload estimated is based on the investment of time in learning and not in teaching. The Curriculum must integrate knowledge of the basic and advanced sciences with problem solving and creativity abilities.

The Curriculum must be broad enough to cover all areas from design to operation of Process plants. It should be deep enough to enable the learners to carry out research and develop products to meet rapidly changing needs and demands. The major challenge in the current scenario is to ensure quality to the stakeholders. Accreditation is the principal means of quality assurance in higher education and reflects the fact that in achieving recognition, the institution or program of study is committed and open to external review to meet certain minimum specified standards. The major emphasis of this accreditation process is to measure the outcomes of the program that is being accredited. Program outcomes are essentially a range of skills and knowledge that a student will have at the time of post-graduation from the program.

With these objectives, online meeting was organized on 7th February 2022 which was attended by heads of the departments and subject faculty of affiliating Institutes. An online meeting was organized on 15th February 2022 with the faculty of MGM CET, Kamothe, Navi Mumbai, teaching the post graduate courses. The program objectives and outcomes were thoroughly discussed in line with AICTE guidelines and the core structure of the syllabus was formulated keeping in mind choice-based credit and grading system curriculum along with more emphasis on learning outcomes. Thus, Skilled based laboratories are introduced in appropriate semesters. Views from experts and PG teachers were taken into consideration and final Academic and Exam scheme was prepared with the consent of all the members involved.

The Program Educational Objectives finalized for the Postgraduate program in Chemical Engineering are:

1. To prepare the student for advancements in mathematical, scientific, engineering excellency and research.
2. To motivate the student to use modern tools for solving real life problems
3. To inculcate a professional and ethical attitude, good leadership qualities and commitment to social and environmental responsibilities.
4. To prepare the student in achieving excellence which will benefit individually and society at large.

Board of Studies in Chemical Engineering

Dr. Parag R Gogte- Chairman

Dr. Kalpana S. Deshmukh - Member

Dr. Sunil J. Kulkarni - Member

Dr. Ramesh S. Bhande – Member

Dr. Shyamala P. Shingare - Member

Dr. Manisha V. Bagal – Member

Dr. Aparna Tamaskar – Member

SEMESTER I

Course Code	CourseName	TeachingScheme(ContactHours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHC101	Advanced Momentum Transfer	3	--	--	3	--	--	3	
CHC102	Advanced Chemical Reaction Engineering	3		--	3		--	3	
CHPE101X	ProgramElective 1	3	--	--	3	--	--	3	
CHPE102X	ProgramElective 2	3	--	--	3	--	--	3	
CHIE101X	InstituteElective1	3	--	--	3	--	--	3	
CHL101	ProgramLab-I	--	2	--	--	1	--	1	
CHSBL101	Skill Based Lab-I	--	4 ^{\$}	--	--	2	--	2	
Total		15	06	--	15	03	--	18	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract /Oral	Total
		InternalAssessment			End Sem.Exam	Exam. Duration (inHrs)			
		Test-1	Test-2	Avg					
CHC101	Advanced Momentum Transfer	20	20	20	80	3	--	--	100
CHC102	Advanced Chemical Reaction Engineering	20	20	20	80	3	--	--	100
CHPE101X	ProgramElective 1	20	20	20	80	3	--	--	100
CHPE102X	ProgramElective 2	20	20	20	80	3	--	--	100
CHIE101X	InstituteElective1	20	20	20	80	3	--	--	100
CHL101	ProgramLab-I	--	--	--	--	--	25	25	50
CHSBL101	Skill Based Lab-I	--	--	--	--	--	50	50	100
Total		--	--	100	400	--	75	75	650

Program Elective 1 (Semester I)

Advanced Heat Transfer (CHPE1011)	Advanced Thermodynamics (CHPE1012)	Process Intensification in Chemical Plants (CHPE1013)
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Program Elective 2 (Semester I)

Corrosion in Industries and its Control (CHPE1021)	Industrial Solid Waste Management (CHPE1022)	Bio Process Engineering(CHPE1023)
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Institute Elective 1 (Semester I)

ProductLifecycleManagement (CHIE1011)	ReliabilityEngineering (CHIE1012)	Management Information System (CHIE1013)
Design of Experiments (CHIE1014)	Operation Research (CHIE1015)	Cyber Security and Laws (CHIE1016)
Disaster Management and Mitigation Measures (CHIE1017)	Energy Audit and Management (CHIE1018)	

SEMESTER II

Course Code	CourseName	TeachingScheme(ContactHours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHC201	Advanced Mass Transfer	3	--	--	3	--	--	3	
CHC202	Advanced Process Control And Dynamics	3		--	3		--	3	
CHPE201X	ProgramElective 3	3	--	--	3	--	--	3	
CHPE202X	ProgramElective 4	3	--	--	3	--	--	3	
CHIE201X	InstituteElective2	3	--	--	3	--	--	3	
CHL201	ProgramLab-II	--	2	--	--	1	--	1	
CHSBL201	SkillBasedLab-II	--	4 ^{\$}	--	--	2	--	2	
Total		15	06	--	15	03	--	18	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract /Oral	Total
		InternalAssessment			End Sem. Exam	Exam. Duration (inHrs)			
		Test-1	Test-2	Avg					
CHC201	Advanced Mass Transfer	20	20	20	80	3	--	--	100
CHC202	Advanced Process Control And Dynamics	20	20	20	80	3	--	--	100
CHPE201X	ProgramElective 3	20	20	20	80	3	--	--	100
CHPE202X	ProgramElective 4	20	20	20	80	3	--	--	100
CHIE201X	InstituteElective2	20	20	20	80	3	--	--	100
CHL201	ProgramLab-II	--	--	--	--	--	25	25	50
CHSBL201	SkillBasedLab-II	--	--	--	--	--	50	50	100
Total		--	--	100	400	--	75	75	650

Program Elective 3 (Semester II)

Advanced Computer Aided Design (CHPE2011)	Heterogeneous Catalysis and Reactor Design (CHPE2012)	Advanced Downstream Processes (CHPE2013)
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Program Elective 4 (Semester II)

Industrial Safety and Hazard Control (CHPE2021)	Green Chemistry and Engineering (CHPE2022)	Industrial Pollution Control and Prevention (CHPE2023)
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Institute Elective 2 (Semester II)

Project Management (CHIE2011)	Finance Management (CHIE2012)	Entrepreneurship Development and Management (CHIE2013)
Human Resource Management (CHIE2014)	Professional Ethics and CSR (CHIE2015)	Digital Business Management (CHIE2016)
Environmental Management (CHIE2017)		

Note 1: Skill Based Lab- I and II are focused on the learning through experience. SBL shall facilitate the learner to acquire the fundamentals of practical engineering in his or her specialization in a project-oriented environment. The learning through skill based labs can be useful in facilitating their research work and hence useful in early completion of their dissertation work.

SEMESTER III

Course Code	CourseName	Teaching Scheme(Contact Hours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHMP301	MajorProject: Dissertation-I	--	20	--	--	10	--	10	
Total		00	20	00	00	10	--	10	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract/ Oral	Total
		InternalAssessment			End Sem. Exam	Exam.D uration(i nHrs)			
		Test-1	Test-2	Avg					
CHMP301	MajorProject: Dissertation-I	--	--	--	--	--	100	--	100
Total		--	--	--	--	--	100	--	100

OnlineCreditCourses

Course Code	CourseName	Teaching Scheme(Contact Hours)			Credits Assigned			
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total
CHOCC301	OnlineCreditCourse-I	--	--	--	--	--	--	3
CHOCC302	OnlineCreditCourse-II	--	--	--	--	--	--	3
Total		--	--	--	00	00	00	06

Note 2: It is mandatory to complete the Online Credit Courses (OCC) available on NPTEL / Swayam /MOOC or similar platform approved by UoM. These two courses shall be completed in any semester I or II or III, but not later end of the Semester III. University shall make a provision that credits earned with OCC- I and OCC- II shall be accounted in the third semester grade-sheet with actual names of courses. The learners shall be allowed to take up these courses from his or her institute or organisation/industry where his/her major project is carried out. The student shall complete the courses and shall qualify the exam conducted by the respective authorities/instructor from the platform. The fees for any such courses and the corresponding examinations shall be borne by the learner.

OnlineCreditCourse –I

The learners shall opt for the course in the domain of Research Methodology or Research & Publication Ethics or IPR. The opted courses shall be of 3 credits of equivalent number of weeks.

OnlineCreditCourse –II

The learners shall opt for the course recommended by Faculty Advisor/Project Supervisor from the institute. The opted course shall be of 3 credits of equivalent number of weeks.

SEMESTER IV

Course Code	CourseName	Teaching Scheme(Contact Hours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHMP401	Major Project :Dissertation-II	--	32	--	--	16	--	16	
Total		--	32	--	--	16	--	16	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract/ Oral	Total
		InternalAssessment			End Sem. Exam	Exam. Duration (inHrs)			
		Test-1	Test-2	Avg					
CHMP401	MajorProject: Dissertation-II	--	--	--	--	--	100	100	200
Total		--	--	--	--	--	100	100	200

Total Credits:68

Note3:TheDissertation-IIsubmissionshallnotbepermittedtillthelearnercompletesalltherequirementsME course.

Note4:Thecontact hoursforthecalculatioofloadofthe teacherforMajorProjectareasfollows:MajorProjectDissertation IandII-02 Hour/ week/student

Semester I		
Course Code	Course Name	Credits
CHC101	Advanced Momentum Transfer	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Physics course involving fluids, Basic concepts of viscosity, stress and strain in fluids.
2. Basic fluid flow course involving equation of continuity, motion and related laminar flow problems.
3. Dimensional Analysis (Buckingham PI theorem), Types of flows, Fluid Statics.

Course Objectives

1. To understand the analogous mechanism of momentum Transport for steady and unsteady flow.
2. To perform momentum balance for a given system at macroscopic and microscopic scale
3. To solve the governing equations to obtain velocity profiles.
4. To model the momentum transport under turbulent conditions
5. To study the mechanism of fluidization and flow through Jet and nozzle.
6. To learn the flow dynamics of compressible fluids.

Detailed Syllabus

Module No.	Course Contents	No.ofHours
1	Shear stress in laminar flow: Newtonian and non-Newtonian fluids; Rheological models; theories of transport properties of gases and liquids; effect of pressure and temperature.	08

	One dimensional momentum transport in laminar flow (shell balance): General method of shell balance approach to momentum transfer problems; momentum flux and velocity distribution for flow of Newtonian and non-Newtonian fluids in pipes, planes, slits and annulus; Fluid flow of two immiscible fluids.	
2	Turbulent flow: basics, Reynolds average Navier-Stokes equations, closure problem, and Boussinesque hypothesis, Prandtl mixing length theory, turbulence models, energy spectrum, turbulent boundary layer, and universal velocity profile.	05
3	Compressible fluids: Flow through variable area- conduits, Flow of gas through a nozzle or orifice (isothermal flow, non isothermal flow), Flow in a pipe (Energy balance for flow of ideal gas, isothermal flow of an ideal gas in a horizontal pipe, Flow with fixed upstream pressure and variable downstream pressure, Non- isothermal flow of an ideal gas in a horizontal pipe, Adiabatic flow of an ideal gas in a horizontal pipe).	08
4	Fluid flow in closed Conduits: Friction factors for fully developed laminar, turbulent and transition flow in circular conduits; Friction factors for flow in the entrance to a circular conduit; Friction factors for packed columns.	03
5	Macroscopic momentum balances: The macroscopic mass, momentum and mechanical energy balances; Use of macroscopic balances for steady-state problems; Use of macroscopic balances for unsteady-state problems.	05
6	Gas-liquid and solid-liquid fluidized beds: Characteristics of particles, Principle of fluidization and mapping of various regimes, Two phase theory of fluidization, Bubbles in fluidized bed, Entrainment and Elutriation, Fast fluidized bed, Mixing, segregation and gas dispersion, Heat and mass transfer in fluidized bed, Solid-liquid fluidized bed and three phase fluidized bed, Design of fluidized bed reactors.	10

Course Outcomes

On completion of the course the students will:

1. Describe origin and importance of Reynolds stresses and estimate values based on simple concepts. Concept of eddy motion scales of turbulence
2. Write Bernoulli's equation and solve simple problems related to flow in pipes, orifice venturi meters, weir, pitot tubes, etc.
3. Calculate friction factor and pressure drop for fluid flow in closed conduits.
4. Calculate minimum fluidization velocity, terminal settling velocity, velocity void age relationships for fluidized beds. Settling velocities under hindered settling conditions.
5. Have knowledge of flow behaviour of compressible fluids.
6. Calculate momentum and mechanical energy balances

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination..
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six question
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four question need to be solved.

Reference Books:

1. Transport Phenomena, R. B. Bird, W. E. Stewart and E.N. Lightfoot, John Wiley, 1960
2. Transport Phenomena – A Unified Approach, R.S. Brodkey and H.C. Hershey, McGraw Hill, 1988
3. Momentum, Heat and Mass transfer, C.O. Bennet and J.E. Myers, McGraw Hill, 1993

Semester I		
Course Code	Course Name	Credits
CHC102	Advanced Chemical Reaction Engineering	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Rate equation of homogeneous and heterogeneous reactions
2. Basic knowledge of different method of analysis of kinetic data.
3. Design equations of batch, semibatch, recycle and flow reactors

Course Objectives

1. To acquire knowledge of non-ideal reactors
2. To study the residence time distribution (RTD) in PFR and CSTR
3. To study rate equations for fluid solid reactions.
4. To understand the concept of heterogeneous reactions.
5. To understand the concept of catalytic reactions.
6. To study different catalytic reactors.

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	RTD for Chemical Reactors General characteristics, Measurement of RTD characteristics, RTD in ideal Reactors, Zero Parameter Models, Segregation Model	06

2	Analysis of Non- Ideal reactors Basic Data, one parameter Models, The tank in the series Models, The dispersion model, two parameter model, Testing the model and determining its parameters, Other models of the non ideal reactors using the CSTR's and PFR's using the RTD	06
3	Rate equation for Fluid- Solid reactions Rate of the absorption, desorption, surface reaction, synthesizing rate law, mechanism and rate limiting steps, design of the reactors for the gas solid reactions, heterogeneous data analysis for the reactor designs, catalysts deactivation, moving bed reactors	06
4	External diffusion effects on the Heterogeneous Reactions Binary diffusion, External resistance to Mass Transfer, The shrinking core model	06
5	Diffusion and Reaction in Porous Catalysts Diffusion and Reactions in spherical catalyst pellets, Internal effectiveness factor, Overall effectiveness factor, Estimation of diffusion and reaction limited regimes, mass transfer and reaction in packed bed, The determination of limiting situation from reaction data	08
6	Design of Heterogeneous Catalytic Reactors Isothermal and adiabatic fixed bed reactors, Non- Isothermal, Non- adiabatic fixed bed reactors, slurry reactors, trickle bed reactors	07

Course Outcomes

On completion of the course the students will:

1. apply knowledge of RTD in design of chemical reactors.
2. analyze and design different non ideal reactors.
3. able to predict which controlling mechanism offer resistance to overall reaction rate.
4. analyze the external diffusion effects on the heterogeneous reactions.
5. able to determine the different properties of catalyst
6. analyze and design heterogeneous catalytic reactors.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks

- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. "Chemical Reaction Engineering" by Octave Levenspiel
2. "Chemical Engineering Kinetics" by J M Smith
3. "Elements of Chemical Reaction Engineering" by H. Scott Fogler

Semester I (Program Elective I)		
Course Code	Course Name	Credits
CHPE1011	Advanced Heat Transfer	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Engineering Mathematics.
2. Fluid Flow.
3. Basic Course on Heat Transfer.
4. Transport Phenomena.

Course Objectives

1. To understand the fundamentals of heat transfer and its applications.
2. To analyze heat transfer in Newtonian and Non-Newtonian fluids.
3. To study heat transfer in industrial process equipment.
4. To familiarize with the design heat transfer equipment such as condensers, reboilers and evaporators.
5. To understand the mechanisms of heat transfer in packed, fluidized and moving bed reactors.
6. To study the design of furnaces and the fundamentals of electrical heating.

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	Overview of Heat Transfer and its Industrial Applications Introduction and importance of the subject. Relevance of subject in industrial environment. Modes of heat transfer. Dissipation of energy in industry. Concept of flow patterns and its uses in industry.	04
2	Heat Transfer in Newtonian and Non-Newtonian Fluids Analogy between momentum and heat transfer. Comparative study of	09

	Newtonian and Non- Newtonian fluid in relation to heat transfer. Newtonian and Non- Newtonian heat transfer in circular tubes, coils and other configurations. Non- Newtonian fluid heat transfer in PFR, CSTR and concept of vibrating / oscillating heat transfer.	
3	Heat Transfer in Industrial Process Equipment Detailed study of Air-Cooled Condensers. Regenerators and Heat Transfer in Agitated Vessels.	06
4	Design of Heat Transfer Equipment Design aspects of Condensers, Reboilers and Evaporators.	08
5	Applications of Heat Transfer in Fluidization Engineering Mechanisms of heat transfer in packed, fluidized and moving bed reactors, heat transfer in dilute phase transport.	05
6	Heat Transfer in Furnaces and Electrical Heating Heat transfer in furnaces, design methods for furnaces, pipe still, thermo siphoning and other industries. Electrical heating.	07

Course Outcomes

On completion of the course the students will be able to:

1. outline the industrial applications of heat transfer.
2. compare heat transfer in Newtonian and Non-Newtonian fluids.
3. analyze heat transfer in industrial process equipment.
4. design heat transfer equipment such as condensers, reboilers and evaporators.
5. predict heat transfer coefficients in packed, fluidized and moving bed reactors.
6. apply the design methods for furnaces and explain electrical heating.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions.
- All questions carry equal marks.
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four questions need to be solved.

Reference Books:

1. Fluid Dynamics and Heat Transfer, J.G. Knudsen and D.L. Katz, McGraw-Hill, 1958.
2. Process Heat Transfer, G.F. Hewitt, G.L. Shires and T.R. Bott, CRC Press, 1994.

3. Fluidization, M. Leva, McGraw-Hill, 1959.
4. Fluidization, J.F. Davidson, R. Clift, D. Harrison, Second Edition, Academic Press, 1985.
5. Fluidization Engineering, D. Kunii and O. Levenspiel, Second Edition, Butterworth-Heinemann, 1991.
6. Non-Newtonian Flow and Heat Transfer, A.H.P. Skelland, Wiley, 1967.
7. Non-Newtonian Flow and Applied Rheology : Engineering Applications, R.P. Chhabra and J.F. Richardson, Second Edition, Elsevier Science, 2008.
8. Heat Transfer to Non-Newtonian Fluids : Fundamentals and Analytical Expressions, A. Shenoy, Wiley-VCH, 2018.
9. Advanced Heat Transfer, G.F. Naterer, Third Edition, CRC Press, 2022.

Semester I (Program Elective I)		
Course Code	Course Name	Credits
CHPE1012	Advanced Thermodynamics	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Basic course in mathematics and thermodynamics
2. Physical Chemistry, kinetic theory of gases,
3. Ideal gas law, vapor pressure and Raoult's law.

Course Objectives

1. To learn advanced topics of classical thermodynamics with emphasis on basic concepts, laws, and thermodynamic relationships
2. To compute thermodynamic properties from volumetric data.
3. To estimate fugacities in gas mixtures based on various cubical equation of states.
4. To analyze various thermodynamic properties in liquid mixtures.
5. To estimate fugacities in liquid mixtures based on various Models and Theories of Solutions
6. To analyze phase equilibrium involving solids

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	Classical Thermodynamics of Phase Equilibria: Gibbs-Duhem equation, chemical potential, fugacity and activity coefficient. Partial Molar Properties.	04
2	Thermodynamic Properties from Volumetric Data: Thermodynamic properties with independent variables P & T, fugacity of a component in a mixture at moderate pressures, thermodynamic properties with independent variables V and	06

	T, fugacity of component in a mixture according to Vander Waal's equation.	
3	Fugacities in Gas mixtures: Lewis fugacity rule, Equation of State, Virial equations of state, Extension to mixtures, fugacities from virial equation, calculation of virial coefficients from potential functions, third virial coefficients, Virial coefficients from corresponding states correlation, fugacities from generalized charts for pure components, fugacities from an empirical equation of state.	08
4	Thermodynamic Properties in Liquid mixtures: Excess functions and fundamental relations of Excess function, activity and activity coefficient, activity coefficients from excess functions in binary mixtures, application of Gibbs-Duhem equation, testing of equilibrium data, Wohl's expansion for excess Gibbs energy, equations of Van der Waal, Wilson and Renon equations, Margules equations, Van Laar equation, UNIQUAC and UNIFAC methods for estimation of activity coefficient. Thermodynamic criteria of miscibility. Azeotropes and their existence.	08
5	Fugacities in Liquid Mixtures- Models and Theories of Solutions: Theory of Van Laar, Scatchard – Hildebrand theory, Lattice theory, Wilson's empirical extension of the Flory–Huggins equation, two liquid theory, chemical theory	06
6	Mixture phase equilibrium involving solids: Solubility of solid in liquid and supercritical fluid, Solid Liquid Equilibrium, Partitioning of solid between two liquid phases, distribution coefficient, Freezing point depression of solvent due to presence of solute, freezing point of liquid mixtures in presence of solid.	07

Course Outcomes

On completion of the course the students will:

1. analyze various thermodynamic properties and their correlations.
2. compute thermodynamic properties of individual species in a mixture.
3. evaluate fugacities in gas mixtures based on various cubical equation of states.
4. estimate activity and activity coefficient in liquid mixtures.
5. evaluate fugacities in liquid mixtures using various Models and Theories of Solutions.
6. analyze phase equilibrium in the presence of solid.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.

- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. J. M. Smith & H. C. Van Ness, "Introduction to Chemical Engineering Thermodynamics"
2. Stanley I. Sandler, "Chemical, Biochemical and Engineering Thermodynamics"
3. Savein Stolen, Tor Grande, Neil Allan, "Chemical Thermodynamics of Materials"
4. K.V.Narayanan, "Chemical Engineering Thermodynamics"
5. Kenneth Denbigh, "Principles of Chemical Equilibrium"
6. Y. V. C. Rao, "Chemical Engineering thermodynamics"
7. B. F. Dodge, "Chemical Engineering Thermodynamics"
8. T. E. Daubert, "Chemical Engineering Thermodynamics"
9. Glasstone S., "Thermodynamics for Chemists"
10. B. G. Kyle, "Chemical and Process Thermodynamics"

Semester I (Program Elective I)		
Course Code	Course Name	Credits
CHPE1013	Process Intensification In Chemical Plants	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

- Basics of Unit Chemical Engineering

Course Objectives

1. To understand the concept of Process Intensification.
2. To provide an understanding of basic operating principles of a variety of intensified process equipments
3. To provide knowledge of application of intensification techniques to a range of processes
4. To understand scale up issues in the chemical process.
5. To gain the scientific background, techniques and applications of intensification in the process industries
6. To Identify and solve process challenges using intensification technologies.

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	Philosophy and principles of process intensification (PI): Introduction, philosophy and opportunities of PI, Types of PI equipments, Equipments and methods. Micro Process Technology in process intensification: Introduction to microprocess technology, Process Intensification by Microreactors, Hydrodynamics and transport in microchannel based microreactor	07
2	High gravity in chemical processing: Historical development, Fundamentals, mechanical design, applications, scale-up and commercial use, future, The spinning disc reactor. Process intensification in extraction: Introduction and Principles, Supercritical extraction for process intensification,	07

3	Multifunctional heat exchanger: Introduction, Compact heat exchanger technology, Single phase flow, Heat transfer and mass transfer, applications.	06
4	Micro reaction technology: Microtechnology, effect of miniaturization, microfabrication, implementation.	05
5	Structured catalysis and reactors: Introduction, overview of structured reactors, Gas phase reactions, multiphase reactions.	05
6	Enhanced fields: Energy based intensifications, Sono-chemistry, Basics of cavitation, Cavitation Reactors, Flow over a rotating surface, Hydrodynamic cavitation applications, Cavitation reactor design, Nusselt-flow model and mass transfer, The Rotating Electrolytic Cell, Microwaves, Electrostatic fields, Sono-crystallization, Reactive separations, Supercritical fluids. Process intensification by membrane: Introduction to membrane and its principles, Membrane engineering in process intensification	09

Course Outcomes

On completion of the course the students will:

1. understand the necessity of Process Intensification
2. apply process intensification in industrial processes
3. implement methodologies for process intensification
4. understand scale up issues in the chemical process
5. know the scientific background, techniques and applications of intensification in the process industries
6. be able to identify and solve process challenges using intensification technologies

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four questions need to be solved.

Reference Books:

1. Reengineering the chemical processing plant, Andrej Stankiewicz, Jacob A., Moulin, Marcel Dekker Inc. New York, Basel.
2. Compact brazed plate heat exchanger. J.M., Navarro, A., Bailly, Elsevier, Paris. 1994.

3. Compact heat exchanger for the process industry. R.K., Shah, Begell House, 1997.
4. Microreactors, Ehrfeld W., Hessel V., Lowe, H., Weinheim: Wiley-VCH, 2000.
5. Conceptual design of chemical processes. J.M., Douglas, McGraw-Hill, New York
6. Reay D., Ramshaw C., Harvey A., Process Intensification, Butterworth Heinemann, 2008.

Semester I (Program Elective II)

Course Code	Course Name	Credits
CHPE1021	Corrosion in industry and its control	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Materials science and Engineering
2. Advanced materials

Course Objectives

1. To understand the knowledge of corrosion prevention methods for designing process industries.
2. To study design aspect for corrosion prevention
3. To design cathodic and anodic protection systems
4. To study corrosion in engineering materials
5. To study corrosion inspection and monitoring
6. To study corrosion and Control methods in industrial environment.

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	Introduction: The economic aspects of corrosion. Case studies. Some examples of catastrophic corrosion failures.	04
2	Design aspect to minimize the corrosion, Temporary corrosion prevention methods and use of inhibitors in industry.	05

3	Design of cathodic and anodic protection systems. Selection of the material for sacrificial anodes – anodes for impressed current methods of protection. Corrosion in pipelines, internal and external protection of pipeline.	08
4	Corrosion of Stainless Steel, Plastics and Elastomers Principal engineering materials for equipment, Corrosion control using the exotic materials. Fabrication of special alloys, Stainless Steel, less common metals, composite materials. Welding techniques and corrosion behavior of weld metals, Plastics and Corrosion of Plastics and Elastomers.	08
5	Corrosion inspection, instrumentation and monitoring, Corrosion testing. Use of DSA in electrochemical industries, Electrochemical Machining, Chemical cleaning of equipment.	07
6	Corrosion and Control methods for water supply systems, cooling systems heavy water systems, underground and marine environments, Biological corrosion. Corrosion nuclear reactors and boilers. Corrosion of reinforcement's concrete structure. Corrosion control in Industrial Environments.	07

Course Outcomes

On completion of the course the students will:

1. have knowledge of economic aspects of corrosion
2. have knowledge of design aspects to minimize the corrosion
3. design of cathodic and anodic protection systems
4. acquire the knowledge of corrosion in different materials
5. apply the knowledge of Corrosion inspection, instrumentation and monitoring
6. acquire the knowledge of Corrosion and Control methods in Industrial Environments

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four questions need to be solved.

Reference Books:

1. M.G.Fontana,CorrosionEngineering,TataMcGraw- Hill(NewDelhi),3rdEd.
2. G.L.ShvartzandM.M.Kristal,CorrosionofChemicalApparatus(1959)ChapmanHallLtd.London

Semester I (Program Elective II)		
Course Code	Course Name	Credits
CHPE1022	Industrial Solid Waste Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Management and engineering for waste disposal.
2. Waste treatment cycles.
3. Basics of unit operations.

Course Objectives

1. To identify key sources, typical qualities generated, composition and properties of solid and hazardous wastes.
2. To identify waste disposal or transformation techniques (landfill).
3. To recognize the relevant, regulations that apply for facilities used for disposal and destruction of waste.
4. To identify and design solid and hazardous waste landfills including closure, post closure and rehab uses.
5. To estimate typical waste disposal costs.
6. To identify recycling and reuse option (composting, source separation and reuse of shredded tyres, recycled glass, fly ash).

Detailed Syllabus

Module No.	Course Contents	Contact Hours
	Introduction: Introduction of solid waste. Classification of solid waste. Importance of solid waste	

1	disposal, engineering principles and management.	06
2	Sources, quantities generated and physicochemical properties of MSW and hazardous waste. Solid waste management pyramid, key technologies for SWM. (Collection, transformation, landfills, composting).	08
3	Types of landfills, basic geotechnical considerations, earthen liners for waste disposal. Clay mineralogy, factors controlling hydraulic conductivity, methods to measure K in the lab and field, compatibility of liner materials to chemicals in leachate.	06
4	Containment and liquid transport in soil liners for RCRA liners (Advection and diffusion). Geosynthesis for waste disposal. Overview, geomembrane leakage, transport and structural stability. Geosynthetic clay liners (GCLS).	06
5	Design of leachate, Collection system for landfills, Use of gravel and GDLs. Operational aspects of MSW landfills (daily cover , leachate disposal , GW monitoring)MSW toenergy (production of biogas).	07
6	System landfill gas collection and leachate recirculation system design. Landfill final cap design and water balance (demonstration of HELP Model) modelling. Review problems.	06

Course Outcomes

On completion of the course the students will:

1. understand engineering principles and management regarding solid waste disposal.
2. understand physicochemical properties of MSW and hazardous waste.
3. determine different factors related to waste disposal.
4. understand different waste disposal techniques.
5. manage economical and environmental regulations for municipal, commercial and industrial solid waste in India and all over the world.
6. apply the knowledge of environmental engineering to convert industrial solid waste into usable energy at minimum operating cost.

Assessment

Internal:

Assessment consists of two tests out of which; one should be a compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.

- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six question
- All question carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four question need to be solved.

Reference Books:

1. Integrated solid waste management by Tehobanoglous/Thusen/Vigil publisher McGraw Hill.
2. Hazardous waste management, 2nd Edition MD LaGrega, PL Buckingham and J.C.Evans McGraw Hill publisher.

Semester I (Program Elective II)		
Course Code	Course Name	Credits
CHPE1023	Bioprocess Engineering	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Application of unit operations in microbes.
2. Alternate of chemical reactions to bioreactions from cost and quality point of view
3. Alternate of synthetic products by natural products

Course Objectives

1. To establish the reaction kinetics model for all types of chemical reaction through bio route.
2. To replace the chemical catalyst by bio-catalyst to reduce the cost of production.
3. To study the cell structure, type of cells, self-fractionation of centrifugation.
4. To study design aspects of bio-reactor versus chemical reactor.
5. To analyze Effect of aeration and agitation during the course of processes.
6. To understand Cell doctrine and biophysics study.

Detailed Syllabus

Module No.	Course Content	No. of Hours
1	Introduction: Definition of BPE and its importance to chemical engineers, industrial operations etc. What is bioprocess engineering, bio-technology and bio-process engineering. Bioprocess regulation constraints. History of Penicillin. How biologists and engineers work together.	05

2	Basic concept of Biology: An engineer's prospective, an overview of biological basics, biophysics and cell doctrine. The structure of all, pro-caryotic cells, Eucaryotic cells, self fractionation, example: analysis of particle motion in any centrifugation. Important cell types, bacteria yeast, moles, algae and protozoa, animal and plant cells, cell construction and cell nutrients.	06
3	Chemicals of life: Lipid, fatty acids and related lipids, fats soluble vitamins steroids and other liquids, sugars and poly saccharides. D glucose and the mono saccharides. Disaccharides to poly saccharides, cellulose and starch, amino acids and proteins. Amino acids building blocks and polypeptides.	06
4	Kinetics of enzyme catalysed reaction: The enzyme substrate complex and enzyme action. Simple enzyme kinetics with one and two substrate. Michaelis-Menten kinetics evaluation of parameters in Michaelis-Menten equation. Kinetics for reversible reaction, two substrate reactions and cofactor activation. Determination of elementary step rate constant, other patterns of substrate concentration dependence, modulation and regulation of enzymatic activity, other influences on enzyme activities, effect of pH and temperature on enzyme and its kinetics. Enzyme deactivation	07
5	Metabolic stoichiometry and energetics: Thermodynamics principles, metabolic reaction coupling, ATP and NAD carbon catabolism, different pathways. Respiration cycles and chains. Biosynthesis of small molecule and macromolecule, transport across cell membrane.	08
6	Kinetics of substrate utilization, product formation and biomass production in cell culture. Ideal reactors for kinetics measurement, i.e. ideal batch CSTR, CSTR in series, FED back plug flow, kinetics of balanced growth, transient growth kinetics, product formation kinetics. Structured kinetic module, product formation kinetics	07

Course Outcomes

On completion of the course the students will:

1. have knowledge of bioprocess engineer, bio-technology and bio-process engineering
2. have knowledge of basic structure of cell and classification of microorganism
3. acquire the knowledge of basic chemicals of life.
4. explain the kinetics of enzyme catalysed reaction in free and immobilized states. They will also be able to organise the production of microbial enzymes and operate variables affecting the production process.
5. acquire the knowledge of Thermodynamics principles, metabolic reaction coupling, ATP and NAD carbon catabolism, different pathways
6. design and analyse bioreactor and design parameters.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

EndSemesterTheoryExamination:

Someguidelinesforsettingupthequestionpaper:

- Minimum80%syllabusshouldbecoveredinquestionpapersofend semesterexamination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in thesyllabus.
- Questionpaperwillcompriseoftotalsixquestion
- Allquestioncarryequalmarks
- Questionswillbemixedinnature(forexamplesupposedQ.2haspart(a)frommodule3the npart(b)willbe fromanymodule otherthanmodule 3)
- OnlyFourquestionneedtobesolved.

Reference Books:

1. Biochemical Engineering Fundamentals by James E Bailey and David F Ollis. Edition 2nd .Tata McGrawHill.
2. BioprocessEngineeringBasic conceptsbyMichael LShuler andFikretKargi.Edition2nd .PHILearningPrivate Limited.

Semester I (Institute Elective I)		
Course Code	Course Name	Credits
CHIE1011	Product Lifecycle Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To familiarize the students with the need, benefits and components of PLM
2. To acquaint students with Product Data Management & PLM strategies
3. To give insights into new product development program and guidelines for designing and developing a product
4. To familiarize the students with Virtual Product Development

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Introduction to Product Lifecycle Management (PLM): Product Lifecycle Management (PLM), Need for PLM, Product Lifecycle Phases, Opportunities of Globalization, Pre-PLM Environment, PLM Paradigm, Importance & Benefits of PLM, Widespread Impact of PLM, Focus and Application, A PLM Project, Starting the PLM Initiative, PLM Applications PLM Strategies: Industrial strategies, Strategy elements, its identification, selection and implementation, Developing PLM Vision and PLM Strategy, Change management for PLM.	10
02	Product Design: Product Design and Development Process, Engineering Design, Organization and Decomposition in Product Design, Typologies of Design Process Models, Reference Model, Product Design in the Context of the Product Development Process, Relation with the Development Process Planning Phase, Relation with the Post design Planning Phase, Methodological Evolution in Product Design, Concurrent Engineering, Characteristic Features of Concurrent Engineering, Concurrent Engineering and Life Cycle Approach, New Product Development (NPD) and Strategies, Product Configuration and Variant Management, The Design for X System, Objective Properties and Design for X Tools, Choice of Design for X Tools and Their Use in the Design Process.	09

03	Product Data Management (PDM): Product and Product Data, PDM systems and importance, Components of PDM, Reason for implementing a PDM system, financial justification of PDM, barriers to PDM implementation.	05
04	Virtual Product Development Tools: For components, machines, and manufacturing plants, 3D CAD systems and realistic rendering techniques, Digital mock-up, Model building, Model analysis, Modeling and simulations in Product Design, Examples/Casestudies.	05
05	Integration of Environmental Aspects in Product Design: Sustainable Development, Design for Environment, Need for Life Cycle Environmental Strategies, Useful Life Extension Strategies, End-of-Life Strategies, Introduction of Environmental Strategies into the Design Process, Life Cycle Environmental Strategies and Considerations for Product Design.	05
06	Life Cycle Assessment and Life Cycle Cost Analysis: Properties, and Framework of Life Cycle Assessment, Phases of LCA in ISO Standards, Fields of Application and Limitations of Life Cycle Assessment, Cost Analysis and the Life Cycle Approach, General Framework for LCCA, Evolution of Models for Product Life Cycle Cost Analysis.	05

Course Outcomes

On completion of the course the students will:

1. Gain knowledge about phases of PLM, PLM strategies and methodology for PLM feasibility study and PDM implementation.
2. Illustrate various approaches and techniques for designing and developing products.
3. Apply product engineering guidelines/thumb rules in designing products for moulding, machining, sheet metal working etc.
4. Acquire knowledge in applying virtual product development tools for components, machining and manufacturing plant.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six question
- All question carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four question need to be solved.

Reference Books:

1. John Stark, "Product Lifecycle Management: Paradigm for 21st Century Product Realisation", Springer- Verlag, 2004. ISBN: 1852338105
2. Fabio Giudice, Guido La Rosa, Antonino Risitano, "Product Design for the environment-A life cycle

- approach”, Taylor & Francis 2006, ISBN: 0849327229
3. Saaksvuori Antti, Immonen Anselmie, “Product Life Cycle Management”, Springer, Dreamtech, ISBN: 3540257314
 4. Michael Grieve, “Product Lifecycle Management: Driving the next generation of lean thinking”, Tata McGraw Hill, 2006, ISBN: 0070636265

Semester I (Institute Elective I)		
Course Code	Course Name	Credits
CHIE1012	Reliability Engineering	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To familiarize the students with various aspects of probability theory
2. To acquaint the students with reliability and its concepts
3. To introduce the students to methods of estimating the system reliability of simple and complex systems
4. To understand the various aspects of Maintainability, Availability and FMEA procedure

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Probability theory: Probability: Standard definitions and concepts; Conditional Probability, Baye's Theorem. Probability Distributions: Central tendency and Dispersion; Binomial, Normal, Poisson, Weibull, Exponential, relations between them and their significance. Measures of Dispersion: Mean Median, Mode, Range, Mean Deviation, Standard Deviation, Variance, Skewness and Kurtosis.	08
02	Reliability Concepts: Reliability definitions, Importance of Reliability, Quality Assurance and Reliability, Bath Tub Curve. Failure Data Analysis: Hazard rate, failure density, Failure Rate, Mean Time To Failure (MTTF), MTBF, Reliability Functions. Reliability Hazard Models: Constant Failure Rate, Linearly increasing, Time Dependent Failure Rate, Weibull Model. Distribution functions and reliability analysis.	08
03	System Reliability: System Configurations: Series, parallel, mixed configuration, k out of n structure, Complex systems.	05
04	Reliability Improvement: Redundancy Techniques: Element redundancy, Unit redundancy, Standby redundancies. Markov analysis. System Reliability Analysis – Enumeration method, Cut-set method, Success Path method, Decomposition method.	08

05	Maintainability and Availability: System downtime, Design for Maintainability: Maintenance requirements, Design methods: Fault Isolation and self-diagnostics, Parts standardization and Interchangeability, Modularization and Accessibility, Repair Vs Replacement. Availability – qualitative aspects.	05
06	Failure Mode, Effects and Criticality Analysis: Failure mode effects analysis, severity/criticality analysis, FMECA examples. Fault tree construction, basic symbols, development of functional reliability block diagram, Fault tree analysis and Event tree Analysis	05

Course Outcomes

On completion of the course the students will:

1. Understand and apply the concept of Probability to engineering problems
2. Apply various reliability concepts to calculate different reliability parameters
3. Estimate the system reliability of simple and complex systems
4. Carry out a Failure Mode Effect and Criticality Analysis

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four questions need to be solved.

References Books:

1. L.S.Srinath, "Reliability Engineering", Affiliated East-West Press (P) Ltd., 1985.
2. Charles E. Ebeling, "Reliability and Maintainability Engineering", Tata McGraw Hill.
3. B.S.Dhillon, C.Singh, "Engineering Reliability", John Wiley & Sons, 1980.
4. P.D.T.Conor, "Practical Reliability Engg.", John Wiley & Sons, 1985.
5. K.C.Kapur, L.R.Lamberson, "Reliability in Engineering Design", John Wiley & Sons.
6. Murray R. Spiegel, "Probability and Statistics", Tata McGraw-Hill Publishing Co. Ltd.

Semester I (Institute Elective I)		
Course Code	Course Name	Credits
CHIE1013	Management Information System	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. The course is blend of Management and Technical field.
2. Discuss the roles played by information technology in today's business and define various technology architectures on which information systems are built
3. Define and analyze typical functional information systems and identify how they meet the need of the firm to deliver efficiency and competitive advantage
4. Identify the basic steps in systems development

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Introduction To Information Systems (IS): Computer Based Information Systems, Impact of IT on organizations, and Importance of IS to Society. Organizational Strategy, Competitive Advantages and IS.	4
02	Data and Knowledge Management: Database Approach, Big Data, Data warehouse and Data Marts, Knowledge Management. Business intelligence (BI): Managers and Decision Making, BI for Data analysis and Presenting Results	7
03	Ethical issues and Privacy: Information Security. Threat to IS, and Security Controls	7
04	Social Computing (SC): Web 2.0 and 3.0, SC in business-shopping, Marketing, Operational and Analytic CRM, E-business and E-commerce – B2B B2C. Mobile commerce.	7
05	Computer Networks Wired and Wireless technology, Pervasive computing, Cloud computing model.	6
06	Information System within Organization: Transaction Processing Systems, Functional Area Information System, ERP and ERP support of Business Process. Acquiring Information Systems and Applications: Various System development lifecycle models.	8

Course Outcomes

On completion of the course the students will:

1. Explain how information systems transform business
2. Identify the impact information systems have on an organization
3. Describe IT infrastructure and its components and its current trends
4. Understand the principal tools and technologies for accessing information from databases to improve business performance and decision making
5. Identify the types of systems used for enterprise-wide knowledge management and how they provide value for businesses

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

References Books:

1. Kelly Rainer, Brad Prince, Management Information Systems, Wiley
2. K.C. Laudon and J.P. Laudon, Management Information Systems: Managing the Digital Firm, 10th Ed., Prentice Hall, 2007.
3. D. Boddy, A. Boonstra, Managing Information Systems: Strategy and Organization, Prentice Hall, 2008

Semester I (Institute Elective I)		
Course Code	Course Name	Credits
CHIE1014	Design of Experiments	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To understand the issues and principles of Design of Experiments (DOE)
2. To list the guidelines for designing experiments
3. To become familiar with methodologies that can be used in conjunction with experimental designs for robustness and optimization

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Introduction 1.1 Strategy of Experimentation 1.2 Typical Applications of Experimental Design 1.3 Guidelines for Designing Experiments 1.4 Response Surface Methodology	06
02	Fitting Regression Models 2.1 Linear Regression Models 2.2 Estimation of the Parameters in Linear Regression Models 2.3 Hypothesis Testing in Multiple Regression 2.4 Confidence Intervals in Multiple Regression 2.5 Prediction of new response observation 2.6 Regression model diagnostics 2.7 Testing for lack of fit	08

03	Two-Level Factorial Designs and Analysis 3.1 The 2^2 Design 3.2 The 2^3 Design 3.3 The General 2^k Design 3.4 A Single Replicate of the 2^k Design 3.5 The Addition of Center Points to the 2^k Design, 3.6 Blocking in the 2^k Factorial Design 3.7 Split-Plot Designs	07
04	Two-Level Fractional Factorial Designs and Analysis 4.1 The One-Half Fraction of the 2^k Design 4.2 The One-Quarter Fraction of the 2^k Design 4.3 The General 2^{k-p} Fractional Factorial Design 4.4 Resolution III Designs 4.5 Resolution IV and V Designs 4.6 Fractional Factorial Split-Plot Designs	07
05	Conducting Tests 5.1 Testing Logistics 5.2 Statistical aspects of conducting tests 5.3 Characteristics of good and bad datasets 5.4 Example experiments 5.5 Attribute Vs Variable datasets	07
06	Taguchi Approach 6.1 Crossed Array Designs and Signal-to-Noise Ratios 6.2 Analysis Methods 6.3 Robust design examples	04

Course Outcomes

On completion of the course the students will:

1. Plan data collection, turn data into information and make decisions that lead to appropriate action
2. Apply the methods taught to real life situations
3. Plan, analyze, and interpret the results of experiments

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)

- Only Four question need to be solved.

References Books:

1. Raymond H. Myers, Douglas C. Montgomery, Christine M. Anderson-Cook, Response Surface Methodology: Process and Product Optimization using Designed Experiment, 3rd edition, John Wiley & Sons, New York, 2001
2. D. C. Montgomery, Design and Analysis of Experiments, 5th edition, John Wiley & Sons, New York, 2001
3. George E. Box, J. Stuart Hunter, William G. Hunter, Statics for Experimenters: Design, Innovation and Discovery, 2nd Ed. Wiley
4. W. J. Dimond, Practical Experiment Designs for Engineers and Scientists, John Wiley and Sons Inc. ISBN: 0-471-39054-2
5. Design and Analysis of Experiments (Springer text in Statistics), Springer by A. M. Dean, and D. T. Voss
6. Phillip J. Ross, "Taguchi Technique for Quality Engineering," McGraw Hill
7. Madhav S. Phadke, "Quality Engineering using Robust Design," Prentice Hall

Semester I (Institute Elective I)		
Course Code	Course Name	Credits
CHIE1015	Operation Research	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. Formulate a real-world problem as a mathematical programming model.
2. Understand the mathematical tools that are needed to solve optimization problems.
3. Use mathematical software to solve the proposed models.

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Introduction to Operations Research: Introduction, Structure of the Mathematical Model, Limitations of Operations Research Linear Programming: Introduction, Linear Programming Problem, Requirements of LPP, Mathematical Formulation of LPP, Graphical method, Simplex Method, Penalty Cost Method or Big M-method, Two Phase Method, Revised simplex method Duality: Primal–Dual construction, Symmetric and Asymmetric Dual, Weak Duality Theorem, Complementary Slackness Theorem, Main Duality Theorem, Dual Simplex Method, Sensitivity Analysis Transportation Problem: Formulation, solution, unbalanced Transportation problem. Finding basic feasible solutions–Northwest corner rule, least cost method and Vogel’s approximation method. Optimality test: the stepping stone method and MODI method. Assignment Problem: Introduction, Mathematical Formulation of the Problem, Hungarian Method Algorithm, Processing of n Jobs Through Two Machines and m Machines, Graphical Method of Two Jobs in Machines Problem Routing Problem, Travelling Salesman Problem Integer Programming Problem: Introduction, Types of Integer Programming Problems, Gomory’s cutting plane Algorithm, Branch and Bound Technique. Introduction to Decompositional algorithms.	14

02	Queuing models: queuing systems and structures, single server and multi-server models, Poisson input, exponential service, constant rate service, finite and infinite population	05
03	Simulation: Introduction, Methodology of Simulation, Basic Concepts, Simulation Procedure, Application of Simulation Monte-Carlo Method: Introduction, Monte-Carlo Simulation, Application of Simulation, Advantages of Simulation, Limitations of Simulation	05
04	Dynamic programming. Characteristics of dynamic programming, Dynamic programming approach for Priority Management, employment smoothing, capital budgeting, Stage Coach/Shortest Path, cargo loading and Reliability problems.	05
05	Game Theory. Competitive games, rectangular game, saddle point, minimax (maximin) method of optimal strategies, value of the game. Solution of games with saddle points, dominance principle. Rectangular games without saddle point – mixed strategy for 2 X 2 games.	05
06	Inventory Models: Classical EOQ Models, EOQ Model with Price Breaks, EOQ with Shortage, Probabilistic EOQ Model,	05

Course Outcomes

On completion of the course the students will:

1. Understand the theoretical workings of the simplex method, the relationship between a linear program and its dual, including strong duality and complementary slackness.
2. Perform sensitivity analysis to determine the direction and magnitude of change of a model's optimal solution as the data change.
3. Solve specialized linear programming problems like the transportation and assignment problems, solve network models like the shortest path, minimum spanning tree, and maximum flow problems.
4. Understand the applications of integer programming and a queuing model and compute important performance measures

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

References Books:

1. Taha, H.A. "Operations Research - An Introduction", Prentice Hall, (7th Edition), 2002.
2. Ravindran, A., Phillips, D. T. and Solberg, J. J. "Operations Research: Principles and Practice", John Wiley and Sons, 2nd Edition, 2009.
3. Hiller, F. S. and Lieberman, G. J. "Introduction to Operations Research", Tata McGraw Hill, 2002.

4. Operations Research, S.D. Sharma, Kedar Nath Ram Nath-Meerut.
 5. Operations Research, Kanti Swarup, P.K. Gupta and Man Mohan, Sultan Chand & Sons.

Semester I (Institute Elective I)		
Course Code	Course Name	Credits
CHIE1016	Cyber Security and Laws	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To understand and identify different types of cybercrime and cyber law
2. To recognize Indian IT Act 2008 and its latest amendments
3. To learn various types of security standards and compliances

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Introduction to Cybercrime: Cybercrime definition and origin of the world, Cybercrime and information security, Classification of cybercrime, Cybercrime and the Indian IT Act 2008, A global perspective on cybercrimes.	4
02	Cyber offenses & Cybercrime: How criminals plan the attacks, Social Engg, Cyber stalking, Cyber café and Cybercrimes, Botnets, Attack vector, Cloud computing, Proliferation of Mobile and Wireless Devices, Trends in Mobility, Credit Card Frauds in Mobile and Wireless Computing Era, Security Challenges Posed by Mobile Devices, Registry Settings for Mobile Devices, Authentication Service Security, Attacks on Mobile/Cell Phones, Mobile Devices: Security Implications for Organizations, Organizational Measures for Handling Mobile Devices-Related Security Issues, Organizational Security Policies and Measures in Mobile Computing Era, Laptops	9
03	Tools and Methods Used in Cyberline Phishing, Password Cracking, Keyloggers and Spywares, Virus and Worms, Steganography, DoS and DDoS Attacks, SQL Injection, Buffer Over Flow, Attacks on Wireless Networks, Phishing, Identity Theft (ID Theft)	6

04	The Concept of Cyberspace E-Commerce , The Contract Aspects in Cyber Law ,The Security Aspect of Cyber Law ,The Intellectual Property Aspect in Cyber Law The Evidence Aspect in Cyber Law ,The Criminal Aspect in Cyber Law,Global Trends in Cyber Law , Legal Framework for Electronic Data Interchange Law Relating to Electronic Banking, The Need for an Indian Cyber Law	8
05	Indian IT Act. Cyber Crime and Criminal Justice: Penalties, Adjudication and Appeals Under the IT Act, 2000, IT Act, 2008 and its Amendments	6
06	Information Security Standard compliances SOX, GLBA, HIPAA, ISO, FISMA, NERC, PCI.	6

Course Outcomes

On completion of the course the students will:

1. Understand the concept of cyber crime and its effect on outside world
2. Interpret and apply IT law in various legal issues
3. Distinguish different aspects of cyber law
4. Apply Information Security Standards compliance during software design and development

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four questions need to be solved.

References Books:

1. Nina Godbole, Sunit Belapure, *Cyber Security*, Wiley India, New Delhi
2. The Indian Cyber Law by Suresh T. Vishwanathan; Bharat Law House New Delhi
3. The Information Technology Act, 2000; Bare Act- Professional Book Publishers, New Delhi.
4. Cyber Law & Cyber Crimes By Advocate Prashant Mali; Snow White Publications, Mumbai
5. Nina Godbole, *Information Systems Security*, Wiley India, New Delhi
6. Kenneth J. Knapp, *Cyber Security & Global Information Assurance* Information Science Publishing.
7. William Stallings, *Cryptography and Network Security*, Pearson Publication
8. Websites for more information are available on: The Information Technology ACT, 2008- TIFR: <https://www.tifrh.res.in>
9. Website for more information, A Compliance Primer for IT professional: <https://www.sans.org/reading->

room/whitepapers/compliance/compliance-primer-professionals-33538

Semester I (Institute Elective I)		
Course Code	Course Name	Credits
CHIE1017	Disaster Management and Mitigation Measures	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To understand physics and various types of disaster occurring around the world
2. To identify extent and damaging capacity of a disaster
3. To study and understand the means of losses and methods to overcome /minimize it.
4. To understand role of individual and various organization during and after disaster
5. To understand application of GIS in the field of disaster management
6. To understand the emergency government response structures before, during and after disaster

.Detailed Syllabus

Module No.	Course Contents	No.of Hours
01	Introduction: Definition of Disaster, hazard, global and Indian scenario, general perspective, importance of study in human life, Direct and indirect effects of disasters, long term effects of disasters. Introduction to global warming and climate change.	03
02	Natural Disaster and Manmade disasters: Natural Disaster: Meaning and nature of natural disaster, Flood, Flash flood, drought, cloudburst, Earthquake, Landslides, Avalanches, Volcanic eruptions, Mudflow, Cyclone, Storm, Storm Surge, climate change, global warming, sea level rise, ozone depletion Manmade Disasters: Chemical, Industrial, Nuclear and Fire Hazards. Role of growing population and subsequent industrialization, urbanization and changing lifestyle of human beings in frequent occurrences of manmade disasters.	09

03	Disaster Management, Policy and Administration: Disaster management: meaning, concept, importance, objective of disaster management policy, disaster risks in India, Paradigm shift in disaster management. Policy and administration: Importance and principles of disaster management policies, command and co-ordination of in disaster management, rescue operations-how to start with and how to proceed in due course of time, study of flowchart showing the entire process.	06
04	Institutional Framework for Disaster Management in India: Importance of public awareness, Preparation and execution of emergency management programme. Scope and responsibilities of National Institute of Disaster Management (NIDM) and National disaster management authority (NDMA) in India. Methods and measures to avoid disasters, Management of casualties, set up of emergency facilities, importance of effective communication among different agencies in such situations. Use of Internet and softwares for effective disaster management. Applications of GIS, Remote sensing and GPS in this regard.	06
05	Financing Relief Measures: Ways to raise finance for relief expenditure, role of government agencies and NGO's in this process, Legal aspects related to finance raising as well as overall management of disasters. Various NGO's and the works they have carried out in the past on the occurrence of various disasters, Ways to approach these teams. International relief aid agencies and their role in extreme events.	09
06	Preventive and Mitigation Measures: Pre-disaster, during disaster and post-disaster measures in some events in general structural mapping: Risk mapping, assessment and analysis, sea walls and embankments, Bio shield, shelters, early warning and communication Non Structural Mitigation: Community based disaster preparedness, risk transfer and risk financing, capacity development and training, awareness and education, contingency plans. Do's and don'ts in case of disasters and effective implementation of relief aids.	06

Course Outcomes

On completion of the course the students will:

1. Get to know natural as well as manmade disaster and their extent and possible effects on the economy.
2. Plan of national importance structures based upon the previous history.
3. Get acquainted with government policies, acts and various organizational structures associated with an emergency.
4. Get to know the simple do's and don'ts in such extreme events and act accordingly.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six question
- All question carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four question need to be solved.

References Books:

1. Disaster Management' by Harsh K. Gupta, Universities Press Publications.
2. 'Disaster Management: An Appraisal of Institutional Mechanisms in India' by O.S. Dagur, published by Centre for land warfare studies, New Delhi, 2011.
3. 'Introduction to International Disaster Management' by Damon Copolla, Butterworth Heinemann Elsevier Publications.
4. 'Disaster Management Handbook' by Jack Pinkowski, CRC Press Taylor and Francis group.
5. 'Disaster management & rehabilitation' by Rajdeep, Dasgupta, Mittal Publications, New Delhi.
6. 'Natural Hazards and Disaster Management, Vulnerability and Mitigation' – RBSingh, Rawat Publications
7. Concepts and Techniques of GIS – C.P. Lo Albert, K.W. Yonng – Prentice Hall (India) Publications. (Learners are expected to refer reports published at national and International level and updated information available on authentic websites)

Semester I (Institute Elective I)		
Course Code	Course Name	Credits
CHIE1018	Energy Audit and Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To understand the importance of energy security for sustainable development and the fundamentals of energy conservation.
2. To introduce performance evaluation criteria of various electrical and thermal installations to facilitate the energy management.
3. To relate the data collected during performance evaluation of systems for identification of energy saving opportunities.

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Energy Scenario: Present Energy Scenario, Energy Pricing, Energy Sector Reforms, Energy Security, Energy Conservation and its Importance, Energy Conservation Act-2001 and its Features. Basics of Energy and its various forms, Material and Energy balance	04
02	Energy Audit Principles: Definition, Energy audit-need, Types of energy audit, Energy management (audit) approach- understanding energy costs, Benchmarking, Energy performance, Matching energy use to requirement, Maximizing system efficiencies, Optimizing the input energy requirements, Fuel and energy substitution. Elements of monitoring & targeting; Energy audit Instruments; Data and information-analysis. Financial analysis techniques: Simple payback period, NPV, Return on investment (ROI), Internal rate of return (IRR)	08
03	Energy Management and Energy Conservation in Electrical System: Electricity billing, Electrical load management and maximum demand Control; Power factor improvement, Energy efficient equipments and appliances, star ratings. Energy efficiency measures in lighting system, Lighting control: Occupancy sensors, daylight integration, and use of intelligent controllers. Energy conservation opportunities in: water pumps, industrial drives, induction motors, motor retrofitting, soft starters, variable speed drives.	10

04	Energy Management and Energy Conservation in Thermal Systems: Review of different thermal loads; Energy conservation opportunities in: Steam distribution system, Assessment of steam distribution losses, Steam leakages, Steam trapping, Condensate and flash steam recovery system. General fuel economy measures in Boilers and furnaces, Waste heat recovery, use of insulation-types and application. HVAC system: Coefficient of performance, Capacity, factors affecting Refrigeration and Air Conditioning system performance and savings opportunities.	10
05	Energy Performance Assessment: On site Performance evaluation techniques, Case studies based on: Motors and variable speed drive, pumps, HVAC system calculations; Lighting System: Installed Load Efficacy Ratio (ILER) method, Financial Analysis.	04
06	Energy conservation in Buildings: Energy Conservation Building Codes (ECBC): Green Building, LEED Rating, Application of Non-Conventional and Renewable Energy Sources	03

Course Outcomes

On completion of the course the students will:

1. To identify and describe present state of energy security and its importance.
2. To identify and describe the basic principles and methodologies adopted in energy audit of an utility.
3. To describe the energy performance evaluation of some common electrical installations and identify the energy saving opportunities.
4. To describe the energy performance evaluation of some common thermal installations and identify the energy saving opportunities.
5. To analyze the data collected during performance evaluation and recommend energy saving measures.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six question
- All question carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four question need to be solved.

References Books:

1. Handbook of Electrical Installation Practice, Geofry Stokes, Blackwell Science
2. Designing with light: Lighting Handbook, By Anil Valia, Lighting System
3. Energy Management Handbook, By W.C. Turner, John Wiley and Sons

4. Handbook on Energy Audits and Management, edited by A.K. Tyagi, Tata Energy Research Institute (TERI).
5. Energy Management Principles, C.B. Smith, Pergamon Press
6. Energy Conservation Guidebook, Dale R. Patrick, S. Fardo, Ray E. Richardson, Fairmont Press
7. Handbook of Energy Audits, Albert Thumann, W.J. Younger, T. Niehus, CRC Press
8. www.energymanagertraining.com
9. www.bee-india.nic.in

Semester I		
Course Code	Course Name	Credits
CHL101	Program Lab-I	01

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
-	02	-	-	01	-	01

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
-	-	-	-	-	25	25	-	50

Prerequisites

- Basic knowledge of reaction kinetics
- Basic knowledge of RTD studies in flow reactor.
- Basic knowledge of Dimensional Analysis Types of flows, Fluid Statics.
- Basic knowledge of involving equation of continuity, motion and related laminar flow

Course Objectives

- To understand the concept of residence time distribution in reactor design
- To study catalytic heterogeneous reaction
- To study the design semibatch reactor
- To Design of fluidized bed reactors
- To study analogous mechanism of momentum Transport for steady and unsteady flow
- To design of stirred tank and different types of mixing equipment

Detailed Syllabus

Program Lab-I should have six experiments based on the courses Advanced Momentum Transfer and Advanced Chemical Reaction Engineering.

Course Outcomes

On completion of the course the students will:

- able to study the residence time distribution in flow reactors
- analyze kinetic data obtain from semi batch reactor

3. able to study the heterogeneous reactions.
4. obtain Friction factors for fully developed laminar, turbulent and transition flow in circular conduits
5. calculate momentum and mechanical energy balances
6. calculate minimum fluidization velocity, terminal setting velocity, velocity void age relationships for fluidized beds. Settling velocities under hindered setting conditions.

Assessment

Laboratory

Weightage for Laboratory should be 40% in Final Assessment of Laboratory Term Work.

Term work

Term work shall be evaluated based on performance in practical.

Practical Journal: 20 marks

Attendance: 05 marks

Total: 25 marks

End Semester Examination

Practical/Oral examination is to be conducted by pair of internal and external examiners

Semester I		
Course Code	Course Name	Credits
CHSBL101	Skill Based Lab-I	02

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
-	04	-	-	02	-	02

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
-	-	-	-	-	50	50	-	100

Prerequisites

1. Basic knowledge Sample Analytical equipment.
2. Practical knowledge of basic engineering subjects like reaction engineering ,fluid dynamics
3. Basic knowledge of basic engineering subjects like reaction engineering ,fluid dynamics
4. Basic knowledge of modeling and simulation

Course Objectives

1. To learn Operation of Analytical Instruments
2. To learn simulation and optimisation of advanced reaction engineering
3. To understand Linear and non linear regression
4. To learn Error analysis methods
5. To learn simulation and optimisation of advanced fluid dynamics
6. To learn the use of process software's used in chemical industries.

Detailed Syllabus

Skill Based Lab-I is focused on the learning through experience. SBL shall facilitate the learner to acquire the fundamentals of practical engineering in his or her specialization in a project-oriented environment. The learning through skill based labs can be useful in facilitating their research work and hence useful in early completion of their dissertation work.

Course Outcomes

On completion of the course the students will:

1. acquire the knowledge to select and use of advanced analytical instruments

2. perform simulation and optimisation of advanced reaction engineering
 3. understand the Linear and non linear regression
 4. perform simulation and optimisation of advanced fluid dynamics
 5. learn error analysis methods.
 6. learn the use of process software's used in chemical industries
- i.

Assessment

Laboratory

Weightage for Laboratory should be 40% in Final Assessment of Laboratory Term Work.

Practical/Oral examination is to be conducted by pair of internal and external examiner.

SEMESTER II

Course Code	CourseName	TeachingScheme(ContactHours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHC201	Advanced Mass Transfer	3	--	--	3	--	--	3	
CHC202	Advanced Process Control And Dynamics	3		--	3		--	3	
CHPE201X	ProgramElective 3	3	--	--	3	--	--	3	
CHPE202X	ProgramElective 4	3	--	--	3	--	--	3	
CHIE201X	InstituteElective2	3	--	--	3	--	--	3	
CHL201	ProgramLab-II	--	2	--	--	1	--	1	
CHSBL201	SkillBasedLab-II	--	4 ^{\$}	--	--	2	--	2	
Total		15	06	--	15	03	--	18	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract /Oral	Total
		InternalAssessment			End Sem. Exam	Exam. Duration (inHrs)			
		Test-1	Test-2	Avg					
CHC201	Advanced Mass Transfer	20	20	20	80	3	--	--	100
CHC202	Advanced Process Control And Dynamics	20	20	20	80	3	--	--	100
CHPE201X	ProgramElective 3	20	20	20	80	3	--	--	100
CHPE202X	ProgramElective 4	20	20	20	80	3	--	--	100
CHIE201X	InstituteElective2	20	20	20	80	3	--	--	100
CHL201	ProgramLab-II	--	--	--	--	--	25	25	50
CHSBL201	SkillBasedLab-II	--	--	--	--	--	50	50	100
Total		--	--	100	400	--	75	75	650

Program Elective 3 (Semester II)

Advanced Computer Aided Design (CHPE2011)	Heterogeneous Catalysis and Reactor Design (CHPE2012)	Advanced Downstream Processes (CHPE2013)
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Program Elective 4 (Semester II)

Industrial Safety and Hazard Control (CHPE2021)	Green Chemistry and Engineering (CHPE2022)	Industrial Pollution Control and Prevention (CHPE2023)
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Institute Elective 2 (Semester II)

Project Management (CHIE2011)	Finance Management (CHIE2012)	Entrepreneurship Development and Management (CHIE2013)
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Human Resource Management (CHIE2014)	Professional Ethics and CSR (CHIE2015)	Digital Business Management (CHIE2016)
Environmental Management (CHIE2017)		

Semester II		
Course Code	Course Name	Credits
CHC201	Advanced Mass Transfer	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Knowledge of chemistry, physics, physical chemistry and mathematics.
2. Knowledge of Diffusion, Film and penetration theories.
3. Knowledge of Distillation, Advanced Topics in distillation.
4. Knowledge of membrane separation process.

Course Objectives

1. To give insight of mass transfer basic principle and mass transfer mechanisms
2. To study the need for reactive separation process
3. To understand design of reactive distillation equipments
4. To understand design of liquid-liquid extraction equipments
5. To understand design of reactive absorption equipments
6. To give insight of membrane separation process.

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1.	Characterization of Separation processes: Inherent Separation Factors: Equilibration Processes, Inherent Separation Factors: Rate-governed Processes. Simple equilibrium processes: Equilibrium Calculations, Checking Phase Conditions for a Mixture.	05
2.	Multistage separation processes: Increasing Product Purity, Reducing Consumption of Separating Agent, Co-current, Crosscurrent, and Countercurrent Flow.	04
3.	Reactive Distillation: Definition, introduction to reactive distillation process. Thermodynamic and kinetic effects on the feasible products of RD: introduction, Azeotropes, azeotropes, kinetics azeotropes in reactive membrane separation, Equilibrium theory and nonlinear waves for reaction separation process. Reactive	09

	stripping in structured catalytic reactors: introduction, hydrodynamics, Reactive experiments, comparison of different internals.	
4.	Liquid-liquid extraction; stage wise calculations for multicomponent with multiple feed stream using reflux and mixed solvents. Liquid-liquid extraction with chemical reaction	07
5.	Reactive Absorption: introduction, reactive absorption equipment, modeling concept, model parameters, case studies. Reactive Extraction: introductions, phase equilibria, reactive mass transfer, hydrodynamics.	07
6.	Membrane Separation Techniques - Basic Equation for membrane separation for permeable & semi permeable membrane, Membrane types & their selection criteria, Technology based Membrane separation like Microfiltration, Ultrafiltration, Reverse Osmosis, Nanofiltration.	07

Course Outcomes

On completion of the course the students will:

1. understand the concept of separation factor and separating agent
2. be able to compare multi-stage operations. design distillation column
3. understand the need for reactive separation process
4. categorize reactive separation processes
5. be able to do sizing of equipment for liquid-liquid extraction, gas absorption
6. understand membrane separation processes principle and working

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. King C. J., Separation Processes, Tata McGraw Hill Book Company, 2nd Ed., New Delhi, 1983.
2. Chemical Engineering Handbook 7th edition by R. H. Perry & Green D.
3. Mass Transfer Operation 3rd Edition by R. E. Treybal
4. Principles of Mass Transfer and Separation Processes, B.K. Dutta
5. Chemical Engineering, Volume 2, J.M. Coulson, J.F. Richardson
6. Transport Processes and Unit Operations, C.J. Geankoplis
7. Transport Processes and Separation Process Principles, C.J. Geankoplis
8. Kulprathipanja, Reactive Separation Processes, Taylor and Francis, 2002.
9. Luyben W. L. and Cheng-Ching Yu, Reactive Distillation Design and Control, John Wiley and Sons, 2008.

Semester II		
Course Code	Course Name	Credits
CHC202	Advanced Process Control and Dynamics	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Mathematics course involving Laplace Transform and its application in solving mathematical equations.
2. Basic course in Linear Open-Loop systems (Response of first and higher-order systems, interacting, non-interacting systems and transportation lag).
3. Linear Closed-Loop Systems (The control system, final control element, Block diagram development of a chemical reactor control system, Transient response of control system, stability, Root-Locus), Frequency response.

Course Objectives

1. To acquire knowledge of designing complex control systems
2. To analyze controller design for processes with difficult dynamics
3. To study controller tuning using approximate process models.
4. To familiarize the design of multi-loop and multivariable control systems.
5. To analyze dynamics of discrete time systems.
6. To understand the principles of model predictive control and design of digital controllers

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	Design of complex control structures : Process with significant disturbance, cascade control, feed forward control, feedback augmented feed forward control, ratio control, processes with multiple outputs controlled by a single input (Override controllers, Auctioneer ring control), Process with single output controlled with multiple input (Split range control, multiple input for improved dynamics), anti-reset windup.	05
2	Controller design for processes with difficult dynamics: Characteristics of difficult process dynamics, non minimum phase system, Time delay system, time delay compensation, inverse response system, inverse response compensation, open loop unstable systems.	06
3	Controller design for non linear systems: Nonlinear controller design philosophies, linearization and classical approach, adaptive control principles (Scheduled adaptive control, model reference adaptive control, self tuning adaptive control), variable transformations.	05
4	Multivariable Regulatory Control Systems: Nature of multivariable systems, multivariable process model, multivariable transfer functions and open loop dynamic analysis, interaction analysis and loop pairing, relative gain array, loop pairing using RGA, loop pairing for nonlinear systems, loop pairing for nonsquare systems, controller design procedure, Decoupling, feasibility of steady state decoupler design, steady state decoupling by singular value decomposition.	07
5	Sampled Data Systems: Sampling and conditioning of continuous signals, signal conditioning, continuous signal reconstruction, mathematical description of discrete-time system, theoretical modeling of discrete time systems. Discrete time system analysis: Basic concepts of z-transforms, inverting z-transforms, Pulse transfer functions, characteristics of open-loop pulse transfer functions, block diagram analysis of sampled data systems, stability.	08
6	Design of digital controllers: The digital controller and its design, discrete PID controller from the continuous domain, other digital controller based on continuous domain strategies, digital Controllers based on discrete domain strategies. Model Predictive Control: General principles of model predictive control, Model algorithmic control, commercial model predictive control schemes, academic and other contributions, nonlinear model predictive control, closing remarks.	08

Course Outcomes

On completion of the course the students will:

1. apply knowledge of process modeling to design complex control structures.
2. design Controllers for processes with difficult dynamics.
3. tune controllers for different controlling processes.
4. analyze and design multi-loop and multivariable control systems.
5. illustrate sampled Data Systems.

6. design digital controllers and analyze Model Predictive control system.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Four questions need to be solved.

Reference Books:

1. "Process Dynamics, Modeling, and Control" by Babatunde A. Ogunnaike, W. Harmon Ray.
2. "Chemical Process Control" by George Stephanopoulos.
3. "Process Control" by Thomas E. Marlin.

Semester II (Program Elective III)		
Course Code	Course Name	Credits
CHPE2011	Advanced Computer Aided Design	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Thermodynamics.
2. Unit Operations.
3. Process Equipment Design.
4. Computer Programming.
5. Modelling and Simulation.

Course Objectives

1. To estimate the physical and thermodynamic properties of process fluids.
2. To study the design of chemical process equipment.
3. To familiarize with the basic structure and components of CAD software.
4. To understand the importance and applications of CAD in the field of Chemical Engineering.
5. To utilize computational software for the design and simulation of chemical process equipment.
6. To design chemical process flow sheets with the help of CAD software.

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	Physical and Thermodynamic Properties of Liquids and Gases Introduction, Estimation of Physical and Thermodynamic Properties of Liquids and Gases, Diffusion Coefficients (Diffusivities), Compressibility Z-Factor of Natural Gases.	06

2	Sizing of Vertical and Horizontal Separators Introduction, Sizing of Vertical and Horizontal Separators, Sizing of Partly Filled Vessels and Tanks.	06
3	Design of Cyclone Separators and Gas Dryers Cyclone Separator Design, Methods for Gas Dehydration, Gas Dryer (Dehydration) Design.	06
4	Mass Transfer Equipment Design Determination of Plates in Fractionating Columns by the Smoker Equations for Binary Mixtures, Multicomponent Distribution and Estimation of Minimum Trays in Distillation Columns.	06
5	Introduction to Computer Software Packages used for Chemical Process Design Types and Characteristics of various Computer Software Packages used for Chemical Process Equipment Design and Flow Sheet Synthesis such as Aspen / Aspen Plus / ChemCad / Hysis (UniSim) / DWSIM.	04
6	Application of Computer Software Packages and Use of Computer Programs for Chemical Process Design Case studies involving the application of Computer Software Packages such as Aspen / Aspen Plus / ChemCad / Hysis (UniSim) / DWSIM and the use of Computer Programs for Design of Chemical Process Equipment and Process Flow Sheet Synthesis.	11

Course Outcomes

On completion of the course the students will be able to:

1. compute the physical and thermodynamic properties of process fluids.
2. calculate the size of Vertical and Horizontal Separators.
3. design Cyclone Separators and Gas Dryers.
4. apply Chemical Engineering Principles in the Design of Mass Transfer Equipment.
5. identify and evaluate various Computer Software Packages for Chemical Process Design.
6. utilize Computer Software Packages and Computer Programs for Design and Simulation of Chemical Process Equipment and Flow Sheet Synthesis.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions.
- All questions carry equal marks.

- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four questions need to be solved.

Reference Books :

1. Coulson and Richardson's Chemical Engineering Vol.6 – Design, R.K. Sinnott (Ed.) Butterworth- Heinemann, New Delhi, 2000.
2. Process Modeling, Simulation and Control for Chemical Engineers, W. Luyben, Second Edition, McGraw-Hill, 1990.
3. Process Modeling and Simulation, R.W. Gaikwad and Dr. Dharendra, Third Edition, Bennet and Co., 2010.
4. Process Simulation and Control using Aspen, A.K. Jana, Prentice Hall of India.
5. Fortran Programs for Chemical Process Design, Analysis and Simulation, A. Kayode Coker, Gulf Publishing Co., 1995.
6. Introduction to Chemical Engineering Computing, B.A. Finlayson, Second Edition, John Wiley and Sons, 2012.
7. Chemical Process Design: Computer-Aided Case Studies, A.C. Dimian and C.S. Bildea, Wiley-VCH, 2008.

Semester II (Program Elective III)		
Course Code	Course Name	Credits
CHPE2012	Heterogeneous Catalysis and Reactor Design	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Rate equation of homogeneous and heterogeneous reactions
2. Basic knowledge of different method of analysis of kinetic data.
3. Design equations of batch, semibatch, recycle and flow reactors

Course Objectives

1. To acquire knowledge of heterogeneous reactions
2. To study the design equations of solid catalyst batch and flow reactors
3. To study Controlling Resistances and the Rate Equation.
4. To understand the design of solid catalysed reactor.
5. To understand the concept of modeling of different types of solid catalysed reactors.
6. To understand the concept of deactivation of catalyst.

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	Heterogeneous Reactions-Introduction, Contacting Patterns for Two-Phase Systems, solid catalysed reactions, Representation of the action of a catalyst. The Spectrum of Kinetic Regimes, The rate equation for surface kinetics.	05

2	Performance equations for reactors containing porous catalyst particles, (Batch reactor, Plug flow reactor and Continuous stirred tank reactor). Experimental methods for finding rates, Comparison of Experimental Reactors. Laboratory Reactors for Accurate Kinetic Data such as Berty Reactor, Carberry Reactor.	06
3	Determining Controlling Resistances and the Rate Equation, Product distribution in multiple reactions, Numericals	05
4	Reactor Design: The Packed Bed Catalytic Reactor, Staged Adiabatic Packed Bed Reactors, Staged Mixed Flow Reactors, Staged Packed Beds with Recycle. Design of single adiabatic packed bed system, Design of a two adiabatic packed bed system.	08
5	Reactors with Suspended Solid Catalyst, Fluidized Reactors of Various Types, background information about suspended solids reactors, The bubbling fluidized bed - BFB, The K-L model for BFB, Application to Catalytic Reactions, First-Order Catalytic Reaction in a BFB, The circulating fluidized bed – CFB, The Fast Fluidized Bed	08
6	Deactivating Catalysts – Mechanisms of catalyst deactivation, The rate and performance equations, Performance Equations in the Regime of Strong Diffusional Resistance, Interpreting kinetic data in the presence of pore diffusion resistance and deactivation, Deactivation in a packed bed reactor.	07

Course Outcomes

On completion of the course the students will:

1. apply knowledge of reaction stoichiometry to reactor design.
2. apply design equations of batch and flow reactors to reactor design.
3. able to predict which controlling mechanism offer resistance to overall reaction rate.
4. analyze and design different types of catalytic reactors.
5. illustrate bubbling fluidized bed and circulating fluidized bed reactor.
6. apply knowledge of deactivation of catalyst.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions

- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. "Chemical Reaction Engineering" by Octave Levenspiel
2. "Chemical Engineering Kinetics" by J M Smith
3. "Elements of Chemical Reaction Engineering" by H. Scott Fogler

Semester II (Program Elective III)		
Course Code	Course Name	Credits
CHPE2013	AdvancedDownstream Processes	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Basic knowledge of Chemical Engineering unit operations.
2. Different separation processes involved in Chemical Engineering
3. Physical Chemistry, kinetic theory of gases
4. Basic knowledge of Process calculations.

Course Objectives

1. Tounderstand the unit processes involved in downstream processing.
2. To acquire knowledge of various advanced downstream processes in Petrochemical industry
3. Tostudy advances in distillation process.
4. Toanalyze energy conseravtion in different separation processes.
5. Tounderstand separation synthesis algorithm
6. Tostudy application of Ion – Exchange techniques in different aspects.

Detailed Syllabus

Module No.	Course Contents	No.ofH ours
1	Introduction: Introduction to Downstream processes theory, applications in chemical separation for Gas-Liquid system, Gas-Solid system. Super critical fluids extraction in food, pharmaceutical, environmental and petroleum applications, water treatment, desalination, Bio separation, dialysis, industrial dialysis.Chromatographic separations specifically Simulated Moving Bed Chromatography for the separation of Racemic Mixtures.	05

2	Downstream Processes in Petrochemical Industry: Cryogenic distillation for refinery, petrochemical off gases, natural gases, gas recovery-Olefin, Helium, Nitrogen, Desulfurization - coal, flue gases.	09
3	Advanced Distillation Processes: Azeotropic & extractive distillation - residue curve maps, homogeneous azeotropic distillation, pressure swing distillation, Column sequences, heterogeneous azeotropic distillation.	10
4	Energy conservation in separation processes: Energy balance, molecular sieves - zeolites, adsorption, catalytic properties, manufacturing processes, hydrogel process, application, New trends.	08
5	Non-Ideal Mixtures: Separations process synthesis for nonazeotropic mixtures, non ideal liquid mixtures, separation synthesis algorithm.	04
6	Ion Exchange: manufacture of resins, physical & chemical properties, capacity, selectivity, application, regeneration, equipment, catalysis use	03

Course Outcomes

On completion of the course the students will:

1. have knowledge of downstream processes involved in various aspects in Chemical industry
2. analyze the application of downstream processes in Petrochemical Industry
3. analyze advanced distillation processes.
4. evaluate energy consumption in different separation processes
5. deal with separation process synthesis for non –ideal mixtures.
6. analyze Ion – Exchange techniques in separation processes.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four questions need to be solved.

Reference Books:

1. Perry's "Chemical Engg. Handbook": McGraw Hill Pub.
2. Douglas J.M., "Conceptual Design of Chemical Processes", McGraw Hill
3. Liu Y.A., "Recent Developments in Chemical Process & Plant Design", John Wiley & Sons Inc.
4. Timmerhaus K.D., "Cryogenic Process Engg.", Plenum Press
5. Othmer Kirk "Encyclopedia of Separation Technology, Vol I & II", Wiley Inter science

Semester II (Program Elective IV)		
Course Code	Course Name	Credits
CHPE2021	Industrial Safety and Hazard Control	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Knowledge of Unit Operations (Heat and Mass Transfer); Basic Chemical Kinetics.
2. General awareness about socio-Industrial culture

Course Objectives

1. To develop awareness about the control measures for Industrial safety and hazards
2. To study various aspects of safety during industrial operations
3. To understand potential hazards of chemical industries.
4. To be acquainted with various rules and regulations related to safety hazards management.
5. To gain knowledge of various treatment and disposal of Hazardous Wastes
6. To gain knowledge of various Safety Management

Detailed Syllabus

Module No.	Course Contents	No.of Hours
1	Introduction: Aspects of Industrial safety, Importance of safety at working places, List of legislations in Chemical industries. The hazardous Wastes (Management and handling) rules, 1989 (2000).	05
2	Chemical Plant Safety: Chemical plant – Fire, Explosion, toxicity, process and corrosion hazards, Plant inspection – check list, Safety audit (IS-14489: 1998), Electrical equipment hazardous area classification, Area classification for instruments, Color codification for identification of pipe lines (IS2379), Indian Standard on safety and health, ILO conventions and recommendations on occupational safety and health.	08

3	Safety Management: The safety management function, Line versus staff authority, Industrial and organized safety, Safety responsibility and accountability, The practice of safety management: The significance of risk acceptability, limitations on standard remedies, safety authority. The problems of holding management's concern: management view of cost, safety and efficiency.	06
4	Hazardous Waste Management: Fundamentals: working definitions, Historical routes, Regulatory initiatives, Classifications, generation, contaminated sites, future endeavors; Legal frame work: Environmental law, Federal hazardous waste Regulations under RCRA, International perspectives.	06
5	Process Fundamentals: A Little chemistry, Physical chemical properties, Energy and mass balance, Reactions and Reactors, Contaminants release, Transportation of contaminants in sub surface; Toxicity: Exposure and toxic effects.	08
6	Treatment and Disposal of Hazardous Wastes: Physical- chemical processes, Biological processes, Stabilization and solidification of waste, Thermal methods- combustion regulations gases and vapors, liquid injection incinerators, solid waste incinerators, storage and feed system, Flue gas temperature reduction,, Air pollution control, Instrumentation, continuous emission monitor, Land disposal/land fill operations.	06

Course Outcomes

On completion of the course, the students will:

1. be well aware of various aspects of industrial safety i.e. safe mode of operating chemical plants, consequences of unsafe operations.
2. have knowledge of potential hazards of various chemicals.
3. be able to deal with hazardous waste management.
4. have knowledge of various methods of disposing hazardous waste.
5. be accustomed with Safety Management
6. have knowledge of treatment and disposal of Hazardous Wastes

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination..
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six question
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four question need to be solved.

Reference

1. "Safety Legislations in Chemical Industries" by S. S. Manakar, Published by Mrs. Jyoti S Manakar, New Panvel

2. “ Safety Management” by John V. Grimaldi, Rollin H. Simonds; 5th Edition, Published by All India traveler Book Seller, Krishan Nagar, Delhi- 110051.
3. “Hazardous Waste Management” by Michael D. LaGrega, Phillips L. Bukingham, Jeffrey C. Evans and Environmental Resource Management, McGraw- Hill International Edition.

Semester II (Program Elective IV)		
Course Code	Course Name	Credits
CHPE2022	Green Chemistry and Engineering	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Engineering Chemistry, Chemical Technology
2. Students should know basic chemistry pertaining to chemical reactions, chemical formula etc.

Course Objectives

1. To understand the principles of Green Chemistry and Engineering.
2. To outline the engineering applications of Green Chemistry and Engineering..
3. To familiarize the process of Green Engineering.
4. To understand the development of environmentally friendly processes.
5. To study the selection of suitable material for the manufacture of industrially important products with minimal formation of waste.
6. To understand methodology for the green process.

Detailed Syllabus

Module No.	Course Contents	No. of Hours
1	Principles of Green Chemistry and Engineering Concept of E Factors and Atom Economy, Principles of Green Chemistry, Waste Minimization, Reduction of Material Use and Energy Requirement, Reduction of Risk and Hazard.	06

2	Green Chemistry and Catalysis Solid Acids and Bases as Catalysts, Catalytic Reductions and Oxidations, Biocatalysis, Phase Transfer Catalysis	06
3	Novel Reaction Media for Green Chemistry and Catalysis Choice of Solvent, Alternative Reaction Media and Multiphasic systems, Supercritical Fluids including Supercritical Carbon Dioxide, Ionic Liquids.	08
4	Chemicals from Renewable Raw Materials Application of Green Chemistry and Engineering in Manufacture of Industrially Important Chemicals such as Ethanol, Lactic Acid, 1,3-Propanediol, Ascorbic Acid, Biodiesel and Green Polymers.	09
5	Process Intensification for Green Processes Spinning Disc Reactors, Microreactors, Intensified Cross-corrugated Multifunctional Membrane, Applications of Ultrasound, Microwaves and Photochemistry for Environmentally Benign Processes, Electrochemistry and Sustainability, Fuel Cells.	08
6	Life-Cycle Assessment for More Sustainable Products and Processes Life-Cycle Assessment (LCA) Methodology, Applications of LCA.	02

Course Outcomes

On completion of the course the students will:

1. apply the principles of Green Chemistry and Engineering to Chemical and Biochemical Processes.
2. understand the phase transfer catalysis process.
3. determine alternative reactions for green chemistry & catalysis.
4. identify industrial based applications of Green Chemistry and Engineering.
5. design and develop processes which are environmentally friendly.
6. produce the desired product with minimal formation of waste.

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six question
- All question carry equal marks
- Questions will be mixed in nature (for example supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)

- Only Four question need to be solved.

Reference Books:

1. Handbook of Green Chemistry and Technology, James Clark and Duncan Macquarrie, Blackwell Science Ltd., 2002.
2. Green Chemistry and Catalysis, Roger Sheldon, Isabel Arends and Ulf Hanefeld, Wiley-VCH Verlag GmbH and Co., 2007.

Semester II (Program Elective IV)		
Course Code	Course Name	Credits
CHPE2023	Industrial Pollution Control and Prevention	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Prerequisites

1. Environmental Engineering and atmospheric pollution.
2. Advanced method of waste treatment.
3. Study and minimization of industrial pollution and its control

Course Objectives

1. To understand the process of industrial pollution.
2. To design the water and air pollution equipment
3. To establish the waste treatment processes as a productive operating plant
4. To analyze the effect of water pollution on human health and establish the standard norms
5. To analyze the effect of water pollution on human health and establish the standard norms
6. To study and analyze pollution control and prevention in specific industries.

.Detailed Syllabus

Module No.	Course Contents	No.of Hours
1	Introduction: Introduction and concept of industrial pollution, what is industrial pollution, how many types of pollution occur in industry explain by industrial examples, why it is important to prevent pollution. Different types of water and air pollution together with noise and nuclear pollution.	06
2	Measurement of industrial pollution: Details of water and air pollution created by different industries, various methods for measurement of water and air pollution in a given industrial sample. Define COD, BOD, DO, TSS, TDS, MLVSS, VSS etc. Determination of industrial pollution norms with their practical aspects.	05
3	Water treatment: Introduction, coagulation, softening, reactors mixing and flocculation, sedimentation, filtration, adsorption, water plant waste management, numerical problems based on industrial wastewater treatment.	06
4	Water quality management and water treatment : Water pollution and their sources in different industries, discharge norms of industrial wastewater, viz. to water bodies. Zero discharge norms for different industries i.e. petroleum, fertilizer, sugar, polymer etc. Recycle of wastewater in industry and its application in prevention of pollution. Wastewater microbiology, characteristics of industrial wastewater, applications of unit operations in primary, secondary and tertiary treatment. Advanced methods of wastewater treatment based on different unit operations. Design of different equipments used for physical, chemical and biological treatment of industrial wastewater.	10
5	Air pollution: Fundamentals of physical and chemical air pollution, air pollutants and their threshold limit, effect of air pollution on environment ultimately on human health, removal of mercury, ammonia, urea, particulate matters, SO ₂ oxide and nitrogen organic vapor from effluent gases.	06
6	Pollution control processes and their economics: Pollution control in selective process industries, pollution control in chemical industries from gaseous considerations, advanced processes which can make the treatment plant a productive plant. Pollution control and its prevention in pulp and paper industry. Pollution control and prevention in miscellaneous process industries. Detailed economics of pollution control from different industries.	06

Course Outcomes

On completion of the course the students will:

1. understand the different types of wastes generated in an industry, their effects on living and non-living things.
2. understand environmental regulatory legislations and standards and climate changes.
3. understand about the quantification and analysis of wastewater and treatment.

4. understand the different unit operations and unit processes involved in conversion of highly polluted water to potable standards.
5. understand about analysis and quantification of hazardous and non-hazardous solid waste wastes, treatment and disposal.
6. have the knowledge of pollution control processes in specific industries.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. "Air pollution" by R.K. Bhatia,
2. "Pollution control in process industry" by S.P. Mahajan.
3. "Introduction to environmental engineering" by David A. Cornwell and Mackenzie L. Davis

Semester II (Institute Elective II)		
Course Code	Course Name	Credits
CHIE2011	Project Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To familiarize the students with the use of a structured methodology/approach for each and every unique project undertaken, including utilizing project management concepts, tools and techniques.
2. To appraise the students with the project management life cycle and make them knowledgeable about the various phases from project initiation through closure.

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Project Management Foundation: Definition of a project, Project Vs Operations, Necessity of project management, Triple constraints, Project life cycles (typical & atypical) Project phases and stage gate process. Role of project manager. Negotiations and resolving conflicts. Project management in various organization structures. PM knowledge areas as per Project Management Institute (PMI).	5
02	Initiating Projects: How to get a project started, Selecting project strategically, Project selection models (Numeric/Scoring Models and Non-numeric models), Project portfolio process, Project sponsor and creating charter; Project proposal. Effective project team, Stages of team development & growth (forming, storming, norming & performing), team dynamics.	6
03	Project Planning and Scheduling: Work Breakdown structure (WBS) and linear responsibility chart, Interface Co-ordination and concurrent engineering, Project cost estimation and budgeting, Top down and bottom up budgeting, Networking and Scheduling techniques. PERT, CPM, GANTT chart. Introduction to Project Management Information System (PMIS).	8

04	Planning Projects: Crashing project time, Resource loading and leveling, Goldratt's critical chain, Project Stakeholders and Communication plan. Risk Management in projects: Risk management planning, Risk identification and risk register. Qualitative and quantitative risk assessment, Probability and impact matrix. Risk response strategies for positive and negative risks	6
05	5.1 Executing Projects: Planning, monitoring and controlling cycle. Information needs and reporting, engaging with all stakeholders of the projects. Team management, communication and project meetings. 5.2 Monitoring and Controlling Projects: Earned Value Management techniques for measuring value of work completed; Using milestones for measurement; change requests and scope creep. Project audit. 5.3 Project Contracting Project procurement management, contracting and outsourcing,	8
06	6.1 Project Leadership and Ethics: Introduction to project leadership, ethics in projects. Multicultural and virtual projects. 6.2 Closing the Project: Customer acceptance; Reasons of project termination, Various types of project terminations (Extinction, Addition, Integration, Starvation), Process of project termination, completing a final report; doing a lessons learned analysis; acknowledging successes and failures; Project management templates and other resources; Managing without authority; Areas of further study	6

Course Outcomes

On completion of the course the students will:

1. Apply selection criteria and select an appropriate project from different options.
2. Write work breakdown structure for a project and develop a schedule based on it.
3. Identify opportunities and threats to the project and decide an approach to deal with them strategically.
4. Use Earned value technique and determine & predict status of the project.
5. Capture lessons learned during project phases and document them for future reference

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)

- Only Four question need to be solved.

Reference Books:

1. Jack Meredith & Samuel Mantel, Project Management: A managerial approach, Wiley India, 7th Ed.
2. A Guide to the Project Management Body of Knowledge (PMBOK[®] Guide), 5th Ed, Project Management Institute PA, USA
3. Gido Clements, Project Management, Cengage Learning.
4. Gopalan, Project Management, Wiley India
5. Dennis Lock, Project Management, Gower Publishing England, 9th Ed.

Semester II (Institute Elective II)		
Course Code	Course Name	Credits
CHIE2012	Finance Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. Overview of Indian financial system, instruments and market
2. Basic concepts of value of money, returns and risks, corporate finance, working capital and its management
3. Knowledge about sources of finance, capital structure, dividend policy

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Overview of Indian Financial System: Characteristics, Components and Functions of Financial System. Financial Instruments: Meaning, Characteristics and Classification of Basic Financial Instruments — Equity Shares, Preference Shares, Bonds-Debentures, Certificates of Deposit, and Treasury Bills. Financial Markets: Meaning, Characteristics and Classification of Financial Markets — Capital Market, Money Market and Foreign Currency Market Financial Institutions: Meaning, Characteristics and Classification of Financial Institutions — Commercial Banks, Investment-Merchant Banks and Stock Exchanges	06
02	Concepts of Returns and Risks: Measurement of Historical Returns and Expected Returns of a Single Security and a Two-security Portfolio; Measurement of Historical Risk and Expected Risk of a Single Security and a Two-security Portfolio. Time Value of Money: Future Value of a Lump Sum, Ordinary Annuity, and Annuity Due; Present Value of a Lump Sum, Ordinary Annuity, and Annuity Due; Continuous Compounding and Continuous Discounting.	06

03	Overview of Corporate Finance: Objectives of Corporate Finance; Functions of Corporate Finance—Investment Decision, Financing Decision, and Dividend Decision. Financial Ratio Analysis: Overview of Financial Statements—Balance Sheet, Profit and Loss Account, and Cash Flow Statement; Purpose of Financial Ratio Analysis; Liquidity Ratios; Efficiency or Activity Ratios; Profitability Ratios; Capital Structure Ratios; Stock Market Ratios; Limitations of Ratio Analysis.	09
04	Capital Budgeting: Meaning and Importance of Capital Budgeting; Inputs for Capital Budgeting Decisions; Investment Appraisal Criterion—Accounting Rate of Return, Payback Period, Discounted Payback Period, Net Present Value (NPV), Profitability Index, Internal Rate of Return (IRR), and Modified Internal Rate of Return (MIRR) Working Capital Management: Concepts of Meaning Working Capital; Importance of Working Capital Management; Factors Affecting an Entity's Working Capital Needs; Estimation of Working Capital Requirements; Management of Inventories; Management of Receivables; and Management of Cash and Marketable Securities.	10
05	Sources of Finance: Long Term Sources—Equity, Debt, and Hybrids; Mezzanine Finance; Sources of Short Term Finance—Trade Credit, Bank Finance, Commercial Paper; Project Finance. Capital Structure: Factors Affecting an Entity's Capital Structure; Overview of Capital Structure Theories and Approaches—Net Income Approach, Net Operating Income Approach; Traditional Approach, and Modigliani-Miller Approach. Relation between Capital Structure and Corporate Value; Concept of Optimal Capital Structure	05
06	Dividend Policy: Meaning and Importance of Dividend Policy; Factors Affecting an Entity's Dividend Decision; Overview of Dividend Policy Theories and Approaches—Gordon's Approach, Walter's Approach, and Modigliani-Miller Approach	03

Course Outcomes

On completion of the course the students will:

1. Understand Indian finance system and corporate finance
2. Take investment, finance as well as dividend decisions

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks

- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. Fundamentals of Financial Management, 13th Edition (2015) by Eugene F. Brigham and Joel F. Houston; Publisher: Cengage Publications, New Delhi.
2. Analysis for Financial Management, 10th Edition (2013) by Robert C. Higgins; Publishers: McGraw Hill Education, New Delhi.
3. Indian Financial System, 9th Edition (2015) by M. Y. Khan; Publisher: McGraw Hill Education, New Delhi.
4. Financial Management, 11th Edition (2015) by I. M. Pandey; Publisher: S. Chand (G/L) & Company Limited, New Delhi

Semester II (Institute Elective II)		
Course Code	Course Name	Credits
CHIE2013	Entrepreneurship Development and Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To acquaint with entrepreneurship and management of business
2. Understand Indian environment for entrepreneurship
3. Idea of EDP, MSME

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Overview Of Entrepreneurship: Definitions, Roles and Functions/Values of Entrepreneurship, History of Entrepreneurship Development, Role of Entrepreneurship in the National Economy, Functions of an Entrepreneur, Entrepreneurship and Forms of Business Ownership, Role of Money and Capital Markets in Entrepreneurial Development: Contribution of Government Agencies in Sourcing information for Entrepreneurship	04
02	Business Plans And Importance Of Capital To Entrepreneurship: Preliminary and Marketing Plans, Management and Personnel, Start-up Costs and Financing as well as Projected Financial Statements, Legal Section, Insurance, Suppliers and Risks, Assumptions and Conclusion, Capital and its Importance to the Entrepreneur Entrepreneurship And Business Development: Starting a New Business, Buying an Existing Business, New Product Development, Business Growth and the Entrepreneur Law and its Relevance to Business Operations	09
03	Women's Entrepreneurship Development, Social entrepreneurship-role and need, EDP cell, role of sustainability and sustainable development for SMEs, case studies, exercises	05

04	Indian Environment for Entrepreneurship: key regulations and legal aspects , MSME Act 2006 and its implications, schemes and policies of the Ministry of MSME, role and responsibilities of various government organisations, departments, banks etc., Role of State governments in terms of infrastructure developments and support etc., Public private partnerships, National Skill development Mission, Credit Guarantee Fund, PMEGP, discussions, group exercises etc	08
05	Effective Management of Business: Issues and problems faced by micro and small enterprises and effective management of MSME enterprises (risk management, credit availability, technology innovation, supply chain management, linkage with large industries), exercises, e-Marketing	08
06	Achieving Success in The Small Business: Stages of the small business life cycle, four types of firm-level growth strategies, Options – harvesting or closing small business Critical Success factors of small business	05

Course Outcomes

On completion of the course the students will:

1. Understand the concept of business plan and ownerships
2. Interpret key regulations and legal aspects of entrepreneurship in India
3. Understand government policies for entrepreneurs

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. Poornima Charantimath, Entrepreneurship development - Small Business Enterprise, Pearson
2. Education Robert D Hisrich, Michael P Peters, Dean A Shepherd, Entrepreneurship, latest edition, The McGraw Hill Company
3. Dr TN Chhabra, Entrepreneurship Development, Sun India Publications, New Delhi
4. Dr CN Prasad, Small and Medium Enterprises in Global Perspective, New Century Publications, New Delhi
5. Vasant Desai, Entrepreneurial development and management, Himalaya Publishing House

6. Maddhurima Lall, Shikah Sahai, Entrepreneurship, Excel Books
7. Rashmi Bansal, STAY hungry STAY foolish, CIIE, IIM Ahmedabad
8. Law and Practice relating to Micro, Small and Medium Enterprises, Taxmann Publication Ltd.
9. Kurakto, Entrepreneurship-Principles and Practices, Thomson Publication
10. Laghu Udyog Samachar
11. www.msme.gov.in
12. www.dcmesme.gov.in
13. www.msme training.gov.in

Semester II (Institute Elective II)		
Course Code	Course Name	Credits
CHIE2014	Human Resource Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To introduce the students with basic concepts, techniques and practices of the human resource management.
2. To provide opportunity of learning Human resource management (HRM) processes, related with the functions, and challenges in the emerging perspective of today's organizations.
3. To familiarize the students about the latest developments, trends & different aspects of HRM.
4. To acquaint the student with the importance of inter-personal & inter-group behavioral skills in an organizational setting required for future stable engineers, leaders and managers.

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Introduction to HR Human Resource Management- Concept, Scope and Importance, Interdisciplinary Approach Relationship with other Sciences, Competencies of HR Manager, HRM functions. Human resource development (HRD): changing role of HRM – Human resource Planning, Technological change, Restructuring and rightsizing, Empowerment, TQM, Managing ethical issues.	5

02	Organizational Behavior (OB) Introduction to OB Origin, Nature and Scope of Organizational Behavior, Relevance to Organizational Effectiveness and Contemporary issues Personality: Meaning and Determinants of Personality, Personality development, Personality Types, Assessment of Personality Traits for Increasing Self Awareness Perception: Attitude and Value, Effect of perception on Individual Decision-making, Attitude and Behavior. Motivation: Theories of Motivation and their Applications for Behavioral Change (Maslow, Herzberg, McGregor); Group Behavior and Group Dynamics: Workgroups formal and informal groups and stages of group development. Team Effectiveness: High performing teams, Team Roles, cross functional and self-directed team. Case study	7
03	Organizational Structure & Design Structure, size, technology, Environment of organization; Organizational Roles & conflicts: Concept of roles; role dynamics; role conflicts and stress. Leadership: Concepts and skills of leadership, Leadership and managerial roles, Leadership styles and contemporary issues in leadership. Power and Politics: Sources and uses of power; Politics at workplace, Tactics and strategies.	6
04	Human Resource Planning Recruitment and Selection process, Job-enrichment, Empowerment-Job-Satisfaction, employee morale. Performance Appraisal Systems: Traditional & modern methods, Performance Counseling, Career Planning. Training & Development: Identification of Training Needs, Training Methods	5
05	Emerging Trends in HR Organizational development; Business Process Re-engineering (BPR), BPR as a tool for organizational development, managing processes & transformation in HR. Organizational Change, Culture, Environment Cross Cultural Leadership and Decision Making: Cross Cultural Communication and diversity at work, Causes of diversity, managing diversity with special reference to handicapped, women and ageing people, intra company cultural difference in employee motivation.	6
06	HR & MIS Need, purpose, objective and role of information system in HR, Applications in HRD in various industries (e.g. manufacturing R&D, Public Transport, Hospitals, Hotels and service industries) Strategic HRM Role of Strategic HRM in the modern business world, Concept of Strategy, Strategic Management Process, Approaches to Strategic Decision Making; Strategic Intent – Corporate Mission, Vision, Objectives and Goals Labor Laws & Industrial Relations Evolution of IR, IR issues in organizations, Overview of Labor Laws in India; Industrial Disputes Act, Trade Unions Act, Shops and Establishments Act	10

Course Outcomes

On completion of the course the students will:

1. Understand the concepts, aspects, techniques and practices of the human resource management.
2. Understand the Human resource management (HRM) processes, functions, changes and challenges in today's emerging organizational perspective.
3. Gain knowledge about the latest developments and trends in HRM.
4. Apply the knowledge of behavioral skills learnt and integrate it within interpersonal and intergroup environment emerging as future stable engineers and managers.

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. Stephen Robbins, Organizational Behavior, 16th Ed, 2013
2. VSP Rao, Human Resource Management, 3rd Ed, 2010, Excel publishing
3. Aswathapa, Human resource management: Text & cases, 6th edition, 2011
4. C.B. Mamoria and SV Gankar, Dynamics of Industrial Relations in India, 15th Ed, 2015, Himalaya Publishing, 15th edition, 2015
5. P. Subba Rao, Essentials of Human Resource management and Industrial relations, 5th Ed, 2013, Himalaya Publishing
6. Laurie Mullins, Management & Organizational Behavior, Latest Ed, 2016, Pearson Publications

Semester II (Institute Elective II)		
Course Code	Course Name	Credits
CHIE2015	Professional Ethics and Corporate Social Responsibility (CSR)	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To understand professional ethics in business
2. To recognize corporate social responsibility

.Detailed Syllabus

Module No.	Course Contents	No. of Hours
01	Professional Ethics and Business: The Nature of Business Ethics; Ethical Issues in Business; Moral Responsibility and Blame; Utilitarianism: Weighing Social Costs and Benefits; Rights and Duties of Business	04
02	Professional Ethics in the Marketplace: Perfect Competition; Monopoly Competition; Oligopolistic Competition; Oligopolies and Public Policy Professional Ethics and the Environment: Dimensions of Pollution and Resource Depletion; Ethics of Pollution Control; Ethics of Conserving Depletable Resources	08
03	Professional Ethics of Consumer Protection: Markets and Consumer Protection; Contract View of Business Firm's Duties to Consumers; Due Care Theory; Advertising Ethics; Consumer Privacy Professional Ethics of Job Discrimination: Nature of Job Discrimination; Extent of Discrimination; Reservation of Jobs.	06
04	Introduction to Corporate Social Responsibility: Potential Business Benefits—Triple bottomline, Human resources, Risk management, Supplier relations; Criticisms and concerns—Nature of business; Motives; Misdirection. Trajectory of Corporate Social Responsibility in India	05
05	Corporate Social Responsibility: Articulation of Gandhian Trusteeship Corporate Social Responsibility and Small and Medium Enterprises (SMEs) in India, Corporate Social Responsibility and Public-Private Partnership (PPP) in India	08

06	Corporate Social Responsibility in Globalizing India: Corporate Social Responsibility Voluntary Guidelines, 2009 issued by the Ministry of Corporate Affairs, Government of India, Legal Aspect of Corporate Social Responsibility—Companies Act, 2013.	08
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Course Outcomes

On completion of the course the students will:

1. Understand rights and duties of business
2. Distinguish different aspects of corporate social responsibility
3. Demonstrate professional ethics
4. Understand legal aspects of corporate social responsibility

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. Business Ethics: Texts and Cases from the Indian Perspective (2013) by Ananda Das Gupta; Publisher: Springer.
2. Corporate Social Responsibility: Readings and Cases in a Global Context (2007) by Andrew Crane, Dirk Matten, Laura Spence; Publisher: Routledge.
3. Business Ethics: Concepts and Cases, 7th Edition (2011) by Manuel G. Velasquez; Publisher: Pearson, New Delhi.
4. Corporate Social Responsibility in India (2015) by Bidyut Chakrabarty, Routledge, New Delhi.

Semester II (Institute Elective II)		
Course Code	Course Name	Credits
CHIE2016	Digital Business Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. To familiarize with digital business concept
2. To acquaint with E-commerce
3. To give insights into E-business and its strategies

.Detailed Syllabus

Module	Detailed content	Contact Hours
1	Introduction to Digital Business- Introduction, Background and current status, E-market places, structures, mechanisms, economics and impacts. Difference between physical economy and digital economy. Drivers of digital business- Big Data & Analytics, Mobile, Cloud Computing, Social media, BYOD, and Internet of Things (digitally intelligent machines/services). Opportunities and Challenges in Digital Business,	09
2	Overview of E-Commerce E-Commerce- Meaning, Retailing, e-commerce- products and services, consumer behavior, market research and advertisement. B2B- E-commerce- selling and buying in private e-markets, public B2B exchanges and support services, e-supply chains, Collaborative Commerce, Intra business EC and Corporate portals. Other E-C models and applications, innovative EC System- From E-government and learning to C2C, mobile commerce and pervasive computing. EC Strategy and Implementation- EC strategy and global EC, Economics and Justification of EC, Using Affiliate marketing to promote your e-commerce business, Launching a successful online business and EC project, Legal, Ethics and Societal impacts of EC.	06

3	Digital Business Support services: ERPase–business backbone, knowledge Tope Apps, Information and referral system Application Development: Building Digital business Applications and Infrastructure	06
4	Managing E-Business- Managing Knowledge, Management skills for e-business, Managing Risks in e-business. Security Threats to e-business- Security Overview, Electronic commerce Threats, Encryption, Cryptography, Public Key and Private Key Cryptography, Digital signatures, Digital Certificates, Security Protocol over Public Networks: HTTP, SSL, Firewall as Security Control, Public Key Infrastructure (PKI) for Security, Prominent Cryptographic Applications	06
5	E-Business Strategy- E-business Strategic formulation- Analysis of Company's Internal and external environment, Selection of strategy, E-business strategy into Action, challenges and E-Transition (Process of Digital Transformation)	04
6	Materializing e-business: From Ideo to Realization- Business plan preparation. Case Studies and presentations	08

Course Outcomes

On completion of the course the students will:

1. Identify drivers of digital business
2. Illustrate various approaches and techniques for E-business and management
3. Prepare E-business plan

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only Four questions need to be solved.

Reference Books:

1. A textbook on E-commerce, Er Arunrajan Mishra, Dr W K Sarwade, Neha Publishers & Distributors, 2011

2. E-commerce from vision to fulfilment, Elias M. Awad, PHI-Restricted, 2002
3. Digital Business and E-Commerce Management, 6th Ed, Dave Chaffey, Pearson, August 2014
4. Introduction to E-business-Management and Strategy, Colin Combe, ELSVIER, 2006
5. Digital Business Concepts and Strategy, Eloise Coupey, 2nd Edition, Pearson
6. Trend and Challenges in Digital Business Innovation, Vinocenzo Morabito, Springer
7. Digital Business Discourse Erika Darics, April 2015, Palgrave Macmillan
8. E-Governance-Challenges and Opportunities in : Proceedings in 2nd International Conference theory and practice of Electronic Governance
9. Perspectives the Digital Enterprise –A framework for Transformation, TCS consulting journal Vol.5
10. Measuring Digital Economy-A new perspective -DOI:10.1787/9789264221796-enOECD Publishing

Semester II (Institute Elective II)		
Course Code	Course Name	Credits
CHIE2017	Environmental Management	03

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
03	-	-	03	-	-	03

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
20	20	20	80	3 Hours	-	-	-	100

Course Objectives

1. Understand and identify environmental issues relevant to India and global concerns
2. Learn concepts of ecology
3. Familiarise environment related legislations

.Detailed Syllabus

Module	Detailed content	Contact Hours
01	Introduction and Definition of Environment: Significance of Environment Management for contemporary managers, Career opportunities. Environmental issues relevant to India, Sustainable Development, and The Energy scenario.	10
02	Global Environmental concerns: Global Warming, Acid Rain, Ozone Depletion, Hazardous Wastes, Endangered life-species, Loss of Biodiversity, Industrial/Man-made disasters, Atomic/Biomedical hazards, etc.	06
03	Concepts of Ecology: Ecosystems and interdependence between living organisms, habitats, limiting factors, carrying capacity, food chain, etc.	05
04	Scope of Environment Management, Role & functions of Government as a planning and regulatory agency. Environment Quality Management and Corporate Environmental Responsibility	10
05	Total Quality Environmental Management, ISO-14000, EMS certification.	05
06	General overview of major legislations like Environment Protection Act, Air (P & CP) Act, Water (P & CP) Act, Wildlife Protection Act, Forest Act, Factories Act, etc.	03

Course Outcomes

On completion of the course the students will:

1. Understand the concept of environmental management
2. Understand ecosystem and interdependence, food chain etc.

3. Understand and interpret environment related legislations

Assessment

Internal:

Assessment consists of two tests out of which; one should be compulsory class test and the other is either a class test or assignment on live problems or course project.

End Semester Theory Examination:

Some guidelines for setting up the question paper:

- Minimum 80% syllabus should be covered in question papers of end semester examination.
- Weightage of each module will be proportional to number of respective lecture hours as mentioned in the syllabus.
- Question paper will comprise of total six questions
- All questions carry equal marks
- Questions will be mixed in nature (for example, supposed Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- Only four questions need to be solved.

Reference Books:

1. Environmental Management: Principles and Practice, C J Barrow, Routledge Publishers London, 1999
2. A Handbook of Environmental Management Edited by Jon C. Lovett and David G. Ockwell, Edward Elgar Publishing
3. Environmental Management, T V Ramachandra and Vijay Kulkarni, TERI Press
4. Indian Standard Environmental Management Systems — Requirements With Guidance For Use, Bureau Of Indian Standards, February 2005
5. Environmental Management: An Indian Perspective, S N Chary and Vinod Vyasulu, Macmillan India, 2000
6. Introduction to Environmental Management, Mary K Theodore and Louise Theodore, CRC Press
7. Environment and Ecology, Majid Hussain, 3rd Ed. Access Publishing, 2015

Semester II		
Course Code	Course Name	Credits
CHL201	Program Lab-II	01

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
-	02	-	-	01	-	01

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
-	-	-	-	-	25	25	-	50

Prerequisites

1. Basic knowledge of all mass transfer operations
2. Practical knowledge of distillation, absorption, adsorption, extraction, leaching, humidification, drying and crystallization
3. Basic knowledge of process modeling, dynamics and control
4. Practical knowledge of process dynamics and feedback control system

Course Objectives

1. To understand basic Equation for membrane separation for membrane
2. To study Experiment on reactive distillation
3. To understand multistage cross current operation in liquid liquid extraction experimentally
4. To analyze cascade, feed forward and ratio control systems
5. To acquire knowledge of tuning PID control system
6. To analyze multi variable control systems

Detailed Syllabus

Program Lab-II should have six experiments based on the courses Advanced Mass Transfer and Advanced Process Control and Dynamics

Course Outcomes

On completion of the course the students will:

1. validate basic Equation for membrane separation for membrane
2. understand reactive distillation operation

3. be able to perform multistage cross current operation in liquid liquid extraction and compare with single stage operation
4. obtain quick control response based on application of advanced control strategies.
5. tune the control parameters for fine system response.
6. achieve pairing of input and output variables in MIMO systems.

Assessment

Laboratory

Weightage for Laboratory should be 40% in Final Assessment of Laboratory Term Work.

Term work

Term work shall be evaluated based on performance in practical.

Practical Journal: 20 marks

Attendance: 05 marks

Total: 25 marks

End Semester Examination

Practical/Oral examination is to be conducted by pair of internal and external examiners

Semester II		
Course Code	Course Name	Credits
CHSBL201	Skill Based Lab-II	02

Course Hours			Credits Assigned			
Theory	Practical	Tutorial	Theory	Practical	Tutorial	Total
-	04	-	-	02	-	02

Theory					Term Work/Practical/Oral			Total
Internal Assessment			End Sem Exam	Duration of End Sem Exam	TW	PR/OR	OR	
Test-I	Test-II	Average						
-	-	-	-	-	50	50	-	100

Prerequisites

1. Basic knowledge Sample Analytical equipment.
2. Practical knowledge of major wastewater analytical test categories.
3. Basic knowledge of process modeling, dynamics and control.
4. Practical knowledge of waste water treatment.

Course Objectives

1. To learn Operation of Analytical Instruments
2. To study Multi component distillation separation and other unit operations using process simulators
3. To understand Linear and non linear regression
4. To learn Error analysis methods
5. To understand Waste Water Analysis
6. To study Development of HEN, Column series

Detailed Syllabus

Skill Based Lab-II is focused on the learning through experience. SBL shall facilitate the learner to acquire the fundamentals of practical engineering in his or her specialization in a project-oriented environment. The learning through skill based labs can be useful in facilitating their research work and hence useful in early completion of their dissertation work.

Course Outcomes

On completion of the course the students will:

1. select the required instruments for sample analysis
2. apply the knowledge of chromatography to Separates the constituents from a complex mixture
3. understand the effects of different constituents in a process outcome and analyse the performance of various on-line and/or off-line instruments and equipment by simulation
4. perform experimental analysis for different offline test TDS, dissolve oxygen etc.
5. learn basic concepts of waste management
6. carryout and determine treatment efficiency of various water treatment processes

Assessment

Laboratory

Weightage for Laboratory should be 40% in Final Assessment of Laboratory Term Work.

Practical/Oral examination is to be conducted by pair of internal and external examiner.

SEMESTER III

Course Code	CourseName	Teaching Scheme(Contact Hours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHMP301	MajorProject: Dissertation-I	--	20	--	--	10	--	10	
Total		00	20	00	00	10	--	10	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract/ Oral	Total
		InternalAssessment			End Sem. Exam	Exam.D uration(i nHrs)			
		Test-1	Test-2	Avg					
CHMP301	MajorProject: Dissertation-I	--	--	--	--	--	100	--	100
Total		--	--	--	--	--	100	--	100

Semester III									
Course Code	CourseName	Teaching Scheme(Contact Hours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHMP301	MajorProject: Dissertation-I	--	20	--	--	10	--	10	
Total		00	20	00	00	10	--	10	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract/ Oral	Total
		InternalAssessment			End Sem. Exam	Exam.D uration(i nHrs)			
		Test-1	Test-2	Avg					
CHMP301	MajorProject: Dissertation-I	--	--	--	--	--	100	--	100
Total		--	--	--	--	--	100	--	100

Course Objectives

1. To investigate the research problems.
2. To acquire critique knowledge from research studies.
3. To identify and formulate the Dissertation Topic with novelty.
4. To compare the related research works in the similar field.
5. To write the literature review scientifically.
6. To develop methodology for the Dissertation topic.

Guidelines for Dissertation-I

Students should do literature survey and identify the problem for Dissertation and finalize in consultation with Guide/Supervisor. Students should use multiple literatures and understand the problem. Students should attempt solution to the problem by analytical/simulation/experimental methods. The solution to be validated with proper justification and compile the report in standard format.

Course Outcomes

On completion of the course the students will:

1. acquire knowledge of various research problems.
2. do critical analysis of upcoming research problems.
3. formulate the Dissertation topic with novelty.
4. do critical analysis of research works in the similar field.
5. have a skill of writing literature review scientifically.

6. finalize the methodology for the Dissertation topic

Guidelines for Assessment of Dissertation-I.

Dissertation-I should be assessed based on following points

- Quality of Literature survey and Novelty in the problem
- Clarity of Problem definition and Feasibility of problem solution
- Relevance to the specialization
- Clarity of objective and scope Dissertation-I should be assessed through a presentation by a panel of Internal examiners and external examiner appointed by the Head of the Department/Institute of respective Programme.

SEMESTER IV

Course Code	CourseName	Teaching Scheme(Contact Hours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHMP401	Major Project :Dissertation-II	--	32	--	--	16	--	16	
Total		--	32	--	--	16	--	16	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract/ Oral	Total
		InternalAssessment			End Sem. Exam	Exam. Duration (inHrs)			
		Test-1	Test-2	Avg					
CHMP401	MajorProject: Dissertation-II	--	--	--	--	--	100	100	200
Total		--	--	--	--	--	100	100	200

Semester IV

Course Code	CourseName	Teaching Scheme(Contact Hours)			Credits Assigned				
		Theory	Pract.	Tut.	Theory	Pract.	Tut.	Total	
CHMP401	Major Project :Dissertation-II	--	32	--	--	16	--	16	
Total		--	32	--	--	16	--	16	
Course Code	CourseName	ExaminationScheme							
		Theory					Term Work	Pract/ Oral	Total
		InternalAssessment			End Sem. Exam	Exam. Duration (inHrs)			
		Test-1	Test-2	Avg					
CHMP401	MajorProject: Dissertation-II	--	--	--	--	--	100	100	200
Total		--	--	--	--	--	100	100	200

Course Objectives

1. To implement the methods/techniques identified for research project.
2. To demonstrate the ability to independently perform primary research activities.
3. To analyze and interpret the results obtained.
4. To develop the ability to compare the results obtained with literature.
5. To demonstrate the original contribution to knowledge.
6. To create a research report.

Guidelines for Dissertation-II

Students should implement the methods/techniques identified in dissertation I. Students should perform primary research activities (experiments, numerical modelling, etc.) on a significant open-ended problem. They should analyze, interpret and compare the result obtained from work and explain thorough and advanced knowledge of research topic in written form by writing a dissertation.

Course Outcomes

On completion of the course the students will:

1. implement the suitable methods/techniques
2. demonstrate the ability to independently perform primary research activities.
3. analyze and interpret the results obtained.
4. compare the results obtained with literature.
5. demonstrate the original contribution to knowledge.
6. elucidate in-depth and advanced knowledge of research topic in written form by writing a dissertation.

GuidelinesforAssessmentofDissertationII

DissertationIIshouldbeassessedbasedonfollowingpoints:

- QualityofLiteraturesurveyand Noveltyin theproblem
- ClarityofProblemdefinitionandFeasibilityofproblemsolution
- Relevancetothespecializationorcurrent Research/ Industrialtrends
- Clarityofobjectiveandscope

- Quality of work attempted or learner contribution
- Validation of results
- Quality of Written and Oral Presentation

Students should publish at least one paper based on the work in referred National/International conference/Journal of repute.

Dissertation II should be assessed by internal and External Examiners appointed by the University of Mumbai.